DECANAL DIMETHYL ACETAL

Other names:	Decylaldehyde dimethylacetal
	Decane, 1,1-dimethoxy-
	1,1-Dimethoxydecane
	Aldehyde C-10 dimethylacetal
	Decylaldehyde dma
	n-Decanal dimethyl acetal
Inchi:	InChI=1S/C12H26O2/c1-4-5-6-7-8-9-10-11-12(13-2)14-3/h12H,4-11H2,1-3H3
InchiKey:	NCRNCSZWOOYBQF-UHFFFAOYSA-N
Formula:	C12H26O2
SMILES:	CCCCCCCCCC(OC)OC
Mol. weight [g/mol]:	202.34

Physical Properties

Property code	Value	Unit	Source
af	0.6079		Cheméo Relay (1.0)
gf	-162.28	kJ/mol	Joback Method
hf	-552.54	kJ/mol	Cheméo Relay (1.0)
hfus	25.69	kJ/mol	Joback Method
hvap	62.49	kJ/mol	Cheméo Relay (1.0)
ie	9.43	eV	Cheméo Relay (1.0)
log10ws	-4.04		Cheméo Relay (1.0)
logp	3.746		Crippen Method
mcvol	191.680	ml/mol	McGowan Method
pc	1741.91	kPa	Joback Method
rinpol	1366.00		NIST Webbook
ripol	1567.00		NIST Webbook
ripol	1567.00		NIST Webbook
tb	502.13	K	Cheméo Relay (1.0)
tc	662.14	K	Cheméo Relay (1.0)
tf	234.16	K	Cheméo Relay (1.0)
vc	0.740	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	464.54	J/mol×K	518.36	Joback Method
cpg	481.08	J/mol×K	545.69	Joback Method
cpg	497.06	J/mol×K	573.02	Joback Method
cpg	512.48	J/mol×K	600.35	Joback Method
cpg	527.34	J/mol×K	627.68	Joback Method
cpg	541.66	J/mol×K	655.01	Joback Method
cpg	555.42	J/mol×K	682.34	Joback Method
dvisc	0.0042647	Pa×s	254.46	Joback Method
dvisc	0.0015728	Pa×s	298.44	Joback Method
dvisc	0.0007494	Pa×s	342.43	Joback Method
dvisc	0.0004228	Pa×s	386.41	Joback Method
dvisc	0.0002681	Pa×s	430.39	Joback Method
dvisc	0.0001850	Pa×s	474.38	Joback Method
dvisc	0.0001359	Pa×s	518.36	Joback Method

Sources

NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C7779411&Units=SI
Joback Method: https://en.wikipedia.org/wiki/Joback_method
McGowan Method: http://link.springer.com/article/10.1007/BF02311772
Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):
Cheméo Relay (1.0, beta):

Legend

af: Acentric Factor
cpg: Ideal gas heat capacity
dvisc: Dynamic viscosity
gf: Standard Gibbs free energy of formation
hf: Enthalpy of formation at standard conditions
hfus: Enthalpy of fusion at standard conditions
hvap: Enthalpy of vaporization at standard conditions
ie: Ionization energy
log10ws: Log10 of Water solubility in mol/l
logp: Octanol/Water partition coefficient

mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume