

# Decanal dimethyl acetal

<b>Other names:</b>	Decylaldehyde dimethylacetal Decane, 1,1-dimethoxy- 1,1-Dimethoxydecane Aldehyde C-10 dimethylacetal Decylaldehyde dma n-Decanal dimethyl acetal
<b>Inchi:</b>	InChI=1S/C12H26O2/c1-4-5-6-7-8-9-10-11-12(13-2)14-3/h12H,4-11H2,1-3H3
<b>InchiKey:</b>	NCRNCSZWOOYBQF-UHFFFAOYSA-N
<b>Formula:</b>	C12H26O2
<b>SMILES:</b>	CCCCCCCCC(OC)OC
<b>Mol. weight [g/mol]:</b>	202.33
<b>CAS:</b>	7779-41-1

## Physical Properties

Property code	Value	Unit	Source
gf	-162.28	kJ/mol	Joback Method
hf	-560.73	kJ/mol	Joback Method
hfus	25.69	kJ/mol	Joback Method
hvap	46.74	kJ/mol	Joback Method
log10ws	-3.63		Crippen Method
logp	3.746		Crippen Method
mcvol	191.680	ml/mol	McGowan Method
pc	1741.91	kPa	Joback Method
rinpol	1366.00		NIST Webbook
ripol	1567.00		NIST Webbook
ripol	1567.00		NIST Webbook
tb	518.36	K	Joback Method
tc	682.34	K	Joback Method
tf	254.46	K	Joback Method
vc	0.738	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	464.54	J/molxK	518.36	Joback Method
cpg	481.08	J/molxK	545.69	Joback Method
cpg	497.06	J/molxK	573.02	Joback Method
cpg	512.48	J/molxK	600.35	Joback Method
cpg	527.34	J/molxK	627.68	Joback Method
cpg	541.66	J/molxK	655.01	Joback Method
cpg	555.42	J/molxK	682.34	Joback Method
dvisc	0.0042647	Paxs	254.46	Joback Method
dvisc	0.0015728	Paxs	298.44	Joback Method
dvisc	0.0007494	Paxs	342.43	Joback Method
dvisc	0.0004228	Paxs	386.41	Joback Method
dvisc	0.0002681	Paxs	430.39	Joback Method
dvisc	0.0001850	Paxs	474.38	Joback Method
dvisc	0.0001359	Paxs	518.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7779411&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7779411&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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