

# 4-Butylbenzoic acid, methyl ester

|                      |                                                                          |
|----------------------|--------------------------------------------------------------------------|
| Other names:         | 4-Butylbenzoic acid, methyl ester<br>methyl 4-butylbenzoate              |
| Inchi:               | InChI=1S/C12H16O2/c1-3-4-5-10-6-8-11(9-7-10)12(13)14-2/h6-9H,3-5H2,1-2H3 |
| InchiKey:            | XJRYQLPELPLPHZ-UHFFFAOYSA-N                                              |
| Formula:             | C12H16O2                                                                 |
| SMILES:              | CCCCc1ccc(C(=O)OC)cc1                                                    |
| Mol. weight [g/mol]: | 192.26                                                                   |

## Physical Properties

| Property code | Value   | Unit    | Source             |
|---------------|---------|---------|--------------------|
| af            | 0.6093  |         | Cheméo Relay (1.0) |
| gf            | -80.98  | kJ/mol  | Joback Method      |
| hf            | -365.47 | kJ/mol  | Cheméo Relay (1.0) |
| hfus          | 23.27   | kJ/mol  | Joback Method      |
| hvap          | 73.82   | kJ/mol  | Cheméo Relay (1.0) |
| ie            | 9.15    | eV      | Cheméo Relay (1.0) |
| log10ws       | -4.10   |         | Cheméo Relay (1.0) |
| logp          | 2.816   |         | Crippen Method     |
| mcvol         | 163.620 | ml/mol  | McGowan Method     |
| pc            | 2480.12 | kPa     | Joback Method      |
| tb            | 537.54  | K       | Cheméo Relay (1.0) |
| tc            | 730.80  | K       | Cheméo Relay (1.0) |
| tf            | 271.40  | K       | Cheméo Relay (1.0) |
| vc            | 0.605   | m3/kmol | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 393.77 | J/mol×K | 581.91          | Joback Method |
| cpg           | 471.60 | J/mol×K | 788.90          | Joback Method |
| cpg           | 460.52 | J/mol×K | 754.40          | Joback Method |
| cpg           | 448.71 | J/mol×K | 719.90          | Joback Method |
| cpg           | 436.14 | J/mol×K | 685.41          | Joback Method |
| cpg           | 422.81 | J/mol×K | 650.91          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 408.69    | J/mol×K | 616.41 | Joback Method |
| dvisc | 0.0004111 | Pa×s    | 459.00 | Joback Method |
| dvisc | 0.0001803 | Pa×s    | 581.91 | Joback Method |
| dvisc | 0.0002277 | Pa×s    | 540.94 | Joback Method |
| dvisc | 0.0002986 | Pa×s    | 499.97 | Joback Method |
| dvisc | 0.0017122 | Pa×s    | 336.10 | Joback Method |
| dvisc | 0.0006026 | Pa×s    | 418.04 | Joback Method |
| dvisc | 0.0009597 | Pa×s    | 377.07 | Joback Method |

## Sources

|                                  |                                                                                                                                               |
|----------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b>           | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>Crippen Method:</b>           | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b>           | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Cheméo Relay (1.0):</b>       |                                                                                                                                               |
| <b>Cheméo Relay (1.0, beta):</b> |                                                                                                                                               |
| <b>NIST Webbook:</b>             | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20651698&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20651698&amp;Units=SI</a> |
| <b>Joback Method:</b>            | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>af:</b>      | Acentric Factor                                 |
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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