

# 1-Butyn-3-one, 1-(6,6-dimethyl-1,2-epoxycyclohexyl)-

Inchi:	InChI=1S/C12H16O2/c1-9(13)6-8-12-10(14-12)5-4-7-11(12,2)3/h10H,4-5,7H2,1-3H3
InchiKey:	ITPNMZLJZQEEOU-UHFFFAOYSA-N
Formula:	C12H16O2
SMILES:	CC(=O)C#CC12OC1CCCC2(C)C
Mol. weight [g/mol]:	192.25

## Physical Properties

Property code	Value	Unit	Source
gf	128.63	kJ/mol	Joback Method
hf	-113.71	kJ/mol	Joback Method
hfus	22.18	kJ/mol	Joback Method
hvap	53.10	kJ/mol	Joback Method
log10ws	-2.78		Crippen Method
logp	1.926		Crippen Method
mcvol	157.060	ml/mol	McGowan Method
pc	3038.96	kPa	Joback Method
rinpol	1462.00		NIST Webbook
rinpol	1462.00		NIST Webbook
tb	577.34	K	Joback Method
tc	821.61	K	Joback Method
tf	483.52	K	Joback Method
vc	0.598	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	404.63	J/mol×K	577.34	Joback Method
cpg	421.77	J/mol×K	618.05	Joback Method
cpg	437.67	J/mol×K	658.76	Joback Method
cpg	452.69	J/mol×K	699.47	Joback Method
cpg	467.14	J/mol×K	740.18	Joback Method
cpg	481.38	J/mol×K	780.89	Joback Method
cpg	495.73	J/mol×K	821.61	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U196972&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U196972&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/85-302-2/1-Butyn-3-one-1-6-6-dimethyl-1-2-epoxycyclohexyl.pdf>

Generated by Cheméo on 2024-04-19 02:14:41.113929321 +0000 UTC m=+15782130.034506642.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.