

# Vanillic acid, n-butyl ester

<b>Other names:</b>	butyl vanillate
<b>Inchi:</b>	InChI=1S/C12H16O4/c1-3-4-7-16-12(14)9-5-6-10(13)11(8-9)15-2/h5-6,8,13H,3-4,7H2,1-12
<b>InchiKey:</b>	TUOSEEPBHUCCAO-UHFFFAOYSA-N
<b>Formula:</b>	C12H16O4
<b>SMILES:</b>	CCCCOC(=O)c1ccc(O)c(OC)c1
<b>Mol. weight [g/mol]:</b>	224.25
<b>CAS:</b>	5348-74-3

## Physical Properties

Property code	Value	Unit	Source
gf	-340.60	kJ/mol	Joback Method
hf	-620.28	kJ/mol	Joback Method
hfus	30.25	kJ/mol	Joback Method
hvap	69.82	kJ/mol	Joback Method
log10ws	-2.64		Crippen Method
logp	2.358		Crippen Method
mcvol	175.360	ml/mol	McGowan Method
pc	2841.41	kPa	Joback Method
tb	684.95	K	Joback Method
tc	899.10	K	Joback Method
tf	470.05	K	Joback Method
vc	0.608	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	472.12	J/molxK	684.95	Joback Method
cpg	485.23	J/molxK	720.64	Joback Method
cpg	497.60	J/molxK	756.33	Joback Method
cpg	509.26	J/molxK	792.03	Joback Method
cpg	520.25	J/molxK	827.72	Joback Method
cpg	530.63	J/molxK	863.41	Joback Method
cpg	540.42	J/molxK	899.10	Joback Method
dvisc	0.0002706	Paxs	470.05	Joback Method

dvisc	0.0001389	Paxs	505.87	Joback Method
dvisc	0.0000779	Paxs	541.68	Joback Method
dvisc	0.0000469	Paxs	577.50	Joback Method
dvisc	0.0000300	Paxs	613.32	Joback Method
dvisc	0.0000201	Paxs	649.13	Joback Method
dvisc	0.0000141	Paxs	684.95	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5348743&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5348743&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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