

# **((CH3)2N)2C=N-(4-CH3-C6H4)**

**Inchi:** InChI=1S/C12H19N3/c1-10-6-8-11(9-7-10)13-12(14(2)3)15(4)5/h6-9H,1-5H3  
**InchiKey:** JQSJEPRNVILZRZ-UHFFFAOYSA-N  
**Formula:** C12H19N3  
**SMILES:** Cc1ccc(N=C(N(C)C)N(C)C)cc1  
**Mol. weight [g/mol]:** 205.30  
**CAS:** 20815-36-5

## **Physical Properties**

<b>Property code</b>	<b>Value</b>	<b>Unit</b>	<b>Source</b>
affp	1044.30	kJ/mol	NIST Webbook
basg	1011.90	kJ/mol	NIST Webbook
hf	141.54	kJ/mol	Joback Method
hvap	52.72	kJ/mol	Joback Method
log10ws	-1.90		Crippen Method
logp	2.106		Crippen Method
mcvol	181.820	ml/mol	McGowan Method
pc	2147.32	kPa	Joback Method
tb	607.06	K	Joback Method
tc	821.87	K	Joback Method

## **Sources**

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C20815365&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## **Legend**

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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