

# Propyl 3-(2-hydroxyphenyl)propanoate

<b>Inchi:</b>	InChI=1S/C12H16O3/c1-2-9-15-12(14)8-7-10-5-3-4-6-11(10)13/h3-6,13H,2,7-9H2,1H3
<b>InchiKey:</b>	WOQCBCOZVWYLKS-UHFFFAOYSA-N
<b>Formula:</b>	C12H16O3
<b>SMILES:</b>	CCCOC(=O)CCc1ccccc1O
<b>Mol. weight [g/mol]:</b>	208.25
<b>CAS:</b>	59116-13-1

## Physical Properties

Property code	Value	Unit	Source
gf	-225.97	kJ/mol	Joback Method
hf	-476.59	kJ/mol	Joback Method
hfus	29.45	kJ/mol	Joback Method
hvap	66.75	kJ/mol	Joback Method
log10ws	-2.36		Crippen Method
logp	2.278		Crippen Method
mcvol	169.490	ml/mol	McGowan Method
pc	2940.89	kPa	Joback Method
rinpol	1598.90		NIST Webbook
tb	657.55	K	Joback Method
tc	873.24	K	Joback Method
tf	435.30	K	Joback Method
vc	0.590	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	448.08	J/molxK	657.55	Joback Method
cpg	461.57	J/molxK	693.50	Joback Method
cpg	474.24	J/molxK	729.45	Joback Method
cpg	486.16	J/molxK	765.40	Joback Method
cpg	497.38	J/molxK	801.34	Joback Method
cpg	507.98	J/molxK	837.29	Joback Method
cpg	518.02	J/molxK	873.24	Joback Method
dvisc	0.0006064	Paxs	435.30	Joback Method

dvisc	0.0002767	Paxs	472.34	Joback Method
dvisc	0.0001415	Paxs	509.38	Joback Method
dvisc	0.0000793	Paxs	546.42	Joback Method
dvisc	0.0000478	Paxs	583.47	Joback Method
dvisc	0.0000306	Paxs	620.51	Joback Method
dvisc	0.0000206	Paxs	657.55	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C59116131&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C59116131&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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