

4-TERT-BUTYL-3-METHOXY-2,6-DINITROTOLUENE

Other names:	1-(1,1-Dimethylethyl)-2-methoxy-4-methyl-3,5-dinitrobenzene 1-Methyl-4-t-butyl-3-methoxy-2,6-dinitrobenzene 1-Methyl-4-tert-butyl-3-methoxy-2,6-dinitrobenzene 2,4-Dinitro-3-methyl-6-tert-butylanisole 2,6-Dinitro-3-methoxy-4-tert-butyltoluene 2-tert-Butyl-4,6-dinitro-5-methylanisole 4-tert-Butyl-2,6-dinitro-3-methoxytoluene 4-tert-Butyl-3-methoxy-2,6-dinitrotoluene 5-t-Butyl-1,3-dinitro-4-methoxy-2-methylbenzene 6-t-Butyl-3-methyl-2,4-dinitroanisole 6-tert-Butyl-3-Methyl-2,4-dinitroanisol 6-tert-Butyl-3-methyl-2,4-dinitroanisole Amber musk Ambrette musk Anisole, 6-t-butyl-3-methyl-2,4-dinitro- Anisole, 6-tert-butyl-3-methyl-2,4-dinitro- Artificial musk ambrette Benzene, 1-(1,1-dimethylethyl)-2-methoxy-4-methyl-3,5-dinitro- Musk amberette Musk ambrette NSC 46122 Synthetic musk ambrette
Inchi:	InChI=1S/C12H16N2O5/c1-7-9(13(15)16)6-8(12(2,3)4)11(19-5)10(7)14(17)18/h6H,1-5H3
InchiKey:	SUAUILGSCPYPJCS-UHFFFAOYSA-N
Formula:	C12H16N2O5
SMILES:	COc1c(C(C)(C)C)cc([N+](=O)[O-])c(C)c1[N+](=O)[O-]
Mol. weight [g/mol]:	268.27

Physical Properties

Property code	Value	Unit	Source
hf	-178.68	kJ/mol	Cheméo Relay (1.0)
hfus	35.82	kJ/mol	Joback Method
hvap	82.16	kJ/mol	Cheméo Relay (1.0)
ie	8.77	eV	Cheméo Relay (1.0)
log10ws	-4.49		Cheméo Relay (1.0)
logp	3.118		Crippen Method

mcvol	196.890	ml/mol	McGowan Method
tf	370.28	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	580.04	J/mol×K	843.43	Joback Method
cpg	592.38	J/mol×K	885.94	Joback Method
cpg	603.61	J/mol×K	928.46	Joback Method
cpg	613.79	J/mol×K	970.97	Joback Method
cpg	622.97	J/mol×K	1013.49	Joback Method
cpg	631.20	J/mol×K	1056.00	Joback Method
cpg	638.53	J/mol×K	1098.52	Joback Method
hsubt	102.90	kJ/mol	323.00	NIST Webbook

Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C83669&Units=SI>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient

mcvol: McGowan's characteristic volume
tf: Normal melting (fusion) point

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