

# 1,1'-Biphenyl 2,3'-dichloro-

<b>Other names:</b>	2,3'-Dichloro-1,1'-biphenyl 2,3'-Dichlorobiphenyl PCB 6
<b>Inchi:</b>	InChI=1S/C12H8Cl2/c13-10-5-3-4-9(8-10)11-6-1-2-7-12(11)14/h1-8H
<b>InchiKey:</b>	ZHBBDTRJIVXKEX-UHFFFAOYSA-N
<b>Formula:</b>	C12H8Cl2
<b>SMILES:</b>	Clc1cccc(-c2cccc2Cl)c1
<b>Mol. weight [g/mol]:</b>	223.10
<b>CAS:</b>	25569-80-6

## Physical Properties

Property code	Value	Unit	Source
gf	231.86	kJ/mol	Joback Method
hf	127.63	kJ/mol	Joback Method
hfus	22.53	kJ/mol	Joback Method
hvap	56.95	kJ/mol	Joback Method
log10ws	-5.44		Crippen Method
logp	4.660		Crippen Method
mcpvol	156.900	ml/mol	McGowan Method
pc	3045.68	kPa	Joback Method
rinpol	1676.00		NIST Webbook
rinpol	1648.90		NIST Webbook
rinpol	1649.00		NIST Webbook
rinpol	1690.00		NIST Webbook
rinpol	1644.00		NIST Webbook
rinpol	1649.00		NIST Webbook
rinpol	1689.00		NIST Webbook
rinpol	1648.90		NIST Webbook
rinpol	1658.00		NIST Webbook
rinpol	1655.00		NIST Webbook
rinpol	1689.00		NIST Webbook
tb	612.14	K	Joback Method
tc	871.88	K	Joback Method
tf	362.72	K	Joback Method
vc	0.590	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	328.07	J/molxK	612.14	Joback Method
cpg	383.35	J/molxK	828.59	Joback Method
cpg	374.26	J/molxK	785.30	Joback Method
cpg	364.26	J/molxK	742.01	Joback Method
cpg	353.28	J/molxK	698.72	Joback Method
cpg	341.24	J/molxK	655.43	Joback Method
cpg	391.60	J/molxK	871.88	Joback Method
dvisc	0.0001963	Paxs	612.14	Joback Method
dvisc	0.0002417	Paxs	570.57	Joback Method
dvisc	0.0003074	Paxs	529.00	Joback Method
dvisc	0.0004074	Paxs	487.43	Joback Method
dvisc	0.0005691	Paxs	445.86	Joback Method
dvisc	0.0008516	Paxs	404.29	Joback Method
dvisc	0.0013975	Paxs	362.72	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25569806&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25569806&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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