

Glutaric acid, 2,2,3,3-tetrafluoropropyl but-3-yn-2-yl ester

Inchi: InChI=1S/C12H14F4O4/c1-3-8(2)20-10(18)6-4-5-9(17)19-7-12(15,16)11(13)14/h1,8,11H
InchiKey: FHPXLBDMGDZCKP-UHFFFAOYSA-N
Formula: C12H14F4O4
SMILES: C#CC(C)OC(=O)CCCC(=O)OCC(F)(F)C(F)F
Mol. weight [g/mol]: 298.23

Physical Properties

Property code	Value	Unit	Source
gf	-975.89	kJ/mol	Joback Method
hf	-1292.46	kJ/mol	Joback Method
hfus	33.24	kJ/mol	Joback Method
hvap	55.14	kJ/mol	Joback Method
log10ws	-3.11		Crippen Method
logp	2.165		Crippen Method
mcvol	193.300	ml/mol	McGowan Method
pc	1932.13	kPa	Joback Method
rinpol	1392.00		NIST Webbook
tb	609.63	K	Joback Method
tc	782.64	K	Joback Method
tf	391.07	K	Joback Method
vc	0.766	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	509.27	J/mol×K	609.63	Joback Method
cpg	521.58	J/mol×K	638.47	Joback Method
cpg	533.24	J/mol×K	667.30	Joback Method
cpg	544.28	J/mol×K	696.14	Joback Method
cpg	554.70	J/mol×K	724.97	Joback Method
cpg	564.53	J/mol×K	753.81	Joback Method
cpg	573.78	J/mol×K	782.64	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U393990&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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