

# Butanedioic acid, methyl phenylmethyl ester

<b>Other names:</b>	Methyl benzyl succinate
<b>Inchi:</b>	InChI=1S/C12H14O4/c1-15-11(13)7-8-12(14)16-9-10-5-3-2-4-6-10/h2-6H,7-9H2,1H3
<b>InchiKey:</b>	MUXGHJRHNFYSBX-UHFFFAOYSA-N
<b>Formula:</b>	C12H14O4
<b>SMILES:</b>	<chem>COC(=O)CCC(=O)OCc1ccccc1</chem>
<b>Mol. weight [g/mol]:</b>	222.24
<b>CAS:</b>	119450-11-2

## Physical Properties

Property code	Value	Unit	Source
gf	-305.27	kJ/mol	Joback Method
hf	-544.08	kJ/mol	Joback Method
hfus	26.45	kJ/mol	Joback Method
hvap	62.89	kJ/mol	Joback Method
log10ws	-2.17		Crippen Method
logp	1.683		Crippen Method
mcvol	171.060	ml/mol	McGowan Method
pc	2637.96	kPa	Joback Method
rinpol	1636.00		NIST Webbook
tb	653.22	K	Joback Method
tc	864.17	K	Joback Method
tf	395.74	K	Joback Method
vc	0.647	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	437.45	J/molxK	653.22	Joback Method
cpg	450.97	J/molxK	688.38	Joback Method
cpg	463.65	J/molxK	723.54	Joback Method
cpg	475.50	J/molxK	758.69	Joback Method
cpg	486.52	J/molxK	793.85	Joback Method
cpg	496.72	J/molxK	829.01	Joback Method
cpg	506.11	J/molxK	864.17	Joback Method

dvisc	0.0013619	Paxs	395.74	Joback Method
dvisc	0.0007885	Paxs	438.65	Joback Method
dvisc	0.0005033	Paxs	481.57	Joback Method
dvisc	0.0003457	Paxs	524.48	Joback Method
dvisc	0.0002514	Paxs	567.39	Joback Method
dvisc	0.0001911	Paxs	610.31	Joback Method
dvisc	0.0001507	Paxs	653.22	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C119450112&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C119450112&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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