

# 2,4-Pentanedione, 3-(phenylmethyl)-

<b>Other names:</b>	2,4-Pentanedione, 3-benzyl- 3-Benzyl-2,4-pentanedione
<b>Inchi:</b>	InChI=1S/C12H14O2/c1-9(13)12(10(2)14)8-11-6-4-3-5-7-11/h3-7,12H,8H2,1-2H3
<b>InchiKey:</b>	WAJQTBOWJRUOOO-UHFFFAOYSA-N
<b>Formula:</b>	C12H14O2
<b>SMILES:</b>	CC(=O)C(Cc1ccccc1)C(C)=O
<b>Mol. weight [g/mol]:</b>	190.24
<b>CAS:</b>	1134-87-8

## Physical Properties

Property code	Value	Unit	Source
gf	-97.71	kJ/mol	Joback Method
hf	-284.92	kJ/mol	Joback Method
hfus	20.55	kJ/mol	Joback Method
hvap	57.69	kJ/mol	Joback Method
log10ws	-2.27		Crippen Method
logp	2.023		Crippen Method
mvol	159.320	ml/mol	McGowan Method
pc	2755.56	kPa	Joback Method
tb	607.94	K	Joback Method
tc	828.22	K	Joback Method
tf	336.28	K	Joback Method
vc	0.606	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	387.83	J/molxK	607.94	Joback Method
cpg	402.29	J/molxK	644.65	Joback Method
cpg	415.78	J/molxK	681.37	Joback Method
cpg	428.35	J/molxK	718.08	Joback Method
cpg	440.03	J/molxK	754.79	Joback Method
cpg	450.87	J/molxK	791.50	Joback Method
cpg	460.91	J/molxK	828.22	Joback Method

dvisc	0.0032069	Paxs	336.28	Joback Method
dvisc	0.0015812	Paxs	381.56	Joback Method
dvisc	0.0009058	Paxs	426.83	Joback Method
dvisc	0.0005774	Paxs	472.11	Joback Method
dvisc	0.0003983	Paxs	517.39	Joback Method
dvisc	0.0002916	Paxs	562.66	Joback Method
dvisc	0.0002237	Paxs	607.94	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1134878&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1134878&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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