

# naphthalenecarboxylic acid, methyl ester

<b>Other names:</b>	methyl naphthoate
<b>Inchi:</b>	InChI=1S/C12H10O2/c1-14-12(13)11-8-4-6-9-5-2-3-7-10(9)11/h2-8H,1H3
<b>InchiKey:</b>	HMRROBKAACRWBP-UHFFFAOYSA-N
<b>Formula:</b>	C12H10O2
<b>SMILES:</b>	COC(=O)c1cccc2ccccc12
<b>Mol. weight [g/mol]:</b>	186.21
<b>CAS:</b>	28804-90-2

## Physical Properties

Property code	Value	Unit	Source
gf	25.67	kJ/mol	Joback Method
hf	-119.68	kJ/mol	Joback Method
hfus	20.29	kJ/mol	Joback Method
hvap	56.04	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	2.626		Crippen Method
mcvol	144.160	ml/mol	McGowan Method
pc	3217.33	kPa	Joback Method
rinpol	1592.00		NIST Webbook
rinpol	279.90		NIST Webbook
rinpol	1592.00		NIST Webbook
tb	600.89	K	Joback Method
tc	836.14	K	Joback Method
tf	368.80	K	Joback Method
vc	0.545	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	336.63	J/molxK	600.89	Joback Method
cpg	349.90	J/molxK	640.10	Joback Method
cpg	362.21	J/molxK	679.31	Joback Method
cpg	373.60	J/molxK	718.51	Joback Method
cpg	384.14	J/molxK	757.72	Joback Method

cpg	393.88	J/molxK	796.93	Joback Method
cpg	402.88	J/molxK	836.14	Joback Method
dvisc	0.0014170	Paxs	368.80	Joback Method
dvisc	0.0009648	Paxs	407.48	Joback Method
dvisc	0.0007022	Paxs	446.16	Joback Method
dvisc	0.0005377	Paxs	484.85	Joback Method
dvisc	0.0004282	Paxs	523.53	Joback Method
dvisc	0.0003519	Paxs	562.21	Joback Method
dvisc	0.0002966	Paxs	600.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C28804902&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C28804902&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/85-570-5/naphthalenecarboxylic-acid-methyl-ester.pdf>

Generated by Cheméo on 2024-04-19 01:15:09.138709303 +0000 UTC m=+15778558.059286619.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.