

Fumaric acid, ethyl 3-nitrophenyl ester

Inchi:	InChI=1S/C12H11NO6/c1-2-18-11(14)6-7-12(15)19-10-5-3-4-9(8-10)13(16)17/h3-8H,2H2
InchiKey:	FLMSXPCTJTUQGF-VOTSOKGWSA-N
Formula:	C12H11NO6
SMILES:	CCOC(=O)C=CC(=O)Oc1cccc([N+](=O)[O-])c1
Mol. weight [g/mol]:	265.22

Physical Properties

Property code	Value	Unit	Source
gf	-199.13	kJ/mol	Joback Method
hf	-449.09	kJ/mol	Joback Method
hfus	37.62	kJ/mol	Joback Method
hvap	80.10	kJ/mol	Joback Method
log10ws	-2.83		Crippen Method
logp	1.619		Crippen Method
mcvol	184.180	ml/mol	McGowan Method
pc	2775.92	kPa	Joback Method
rinpol	2224.00		NIST Webbook
rinpol	2224.00		NIST Webbook
tb	814.20	K	Joback Method
tc	1053.13	K	Joback Method
tf	546.79	K	Joback Method
vc	0.710	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	506.62	J/molxK	814.20	Joback Method
cpg	517.16	J/molxK	854.02	Joback Method
cpg	526.71	J/molxK	893.84	Joback Method
cpg	535.30	J/molxK	933.66	Joback Method
cpg	542.95	J/molxK	973.49	Joback Method
cpg	549.70	J/molxK	1013.31	Joback Method
cpg	555.56	J/molxK	1053.13	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U348185&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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