

1-Methoxycarbonylmethyl-cyclopropane-1,2-dicarboxylic acid dimethyl ester

Inchi: COC(=O)CC1(C(=O)OC)CC1C(=O)OC
InchiKey: YGEPYQNFCTTQQQ-UHFFFAOYSA-N

Formula: C10H14O6
SMILES: COC(=O)CC1(C(=O)OC)CC1C(=O)OC
Mol. weight [g/mol]: 230.21

Physical Properties

Property code	Value	Unit	Source
gf	-620.89	kJ/mol	Joback Method
hf	-916.43	kJ/mol	Joback Method
hfus	22.93	kJ/mol	Joback Method
hvap	63.78	kJ/mol	Joback Method
log10ws	-7.47e-03		Crippen Method
logp	-0.098		Crippen Method
mcvol	163.220	ml/mol	McGowan Method
pc	2775.92	kPa	Joback Method
rinpol	1428.00		NIST Webbook
tb	659.38	K	Joback Method
tc	863.43	K	Joback Method
tf	456.54	K	Joback Method
vc	0.622	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	442.73	J/mol×K	659.38	Joback Method
cpg	455.23	J/mol×K	693.39	Joback Method
cpg	467.18	J/mol×K	727.40	Joback Method
cpg	478.67	J/mol×K	761.40	Joback Method
cpg	489.74	J/mol×K	795.41	Joback Method
cpg	500.47	J/mol×K	829.42	Joback Method
cpg	510.93	J/mol×K	863.43	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R249122&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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