

# Succinic acid, di(but-2-en-1-yl) ester

<b>Inchi:</b>	InChI=1S/C12H18O4/c1-3-5-9-15-11(13)7-8-12(14)16-10-6-4-2/h3-6H,7-10H2,1-2H3/b5-
<b>InchiKey:</b>	PUASBHVEXSYHGK-GGWOSOGESA-N
<b>Formula:</b>	C12H18O4
<b>SMILES:</b>	CC=CCOC(=O)CCC(=O)OCC=CC
<b>Mol. weight [g/mol]:</b>	226.27

## Physical Properties

Property code	Value	Unit	Source
gf	-257.24	kJ/mol	Joback Method
hf	-546.17	kJ/mol	Joback Method
hfus	32.81	kJ/mol	Joback Method
hvap	60.53	kJ/mol	Joback Method
log10ws	-2.28		Crippen Method
logp	2.005		Crippen Method
mcvol	186.220	ml/mol	McGowan Method
pc	2115.83	kPa	Joback Method
rinpol	1577.00		NIST Webbook
tb	634.86	K	Joback Method
tc	825.03	K	Joback Method
tf	359.16	K	Joback Method
vc	0.716	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	473.85	J/molxK	634.86	Joback Method
cpg	487.41	J/molxK	666.55	Joback Method
cpg	500.28	J/molxK	698.25	Joback Method
cpg	512.49	J/molxK	729.94	Joback Method
cpg	524.06	J/molxK	761.64	Joback Method
cpg	535.00	J/molxK	793.33	Joback Method
cpg	545.33	J/molxK	825.03	Joback Method
dvisc	0.0014657	Paxs	359.16	Joback Method
dvisc	0.0007481	Paxs	405.11	Joback Method

dvisc	0.0004380	Paxs	451.06	Joback Method
dvisc	0.0002831	Paxs	497.01	Joback Method
dvisc	0.0001970	Paxs	542.96	Joback Method
dvisc	0.0001451	Paxs	588.91	Joback Method
dvisc	0.0001117	Paxs	634.86	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391233&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391233&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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