

# 2-Butenoic acid, isohexyl ester

|                      |                                                                           |
|----------------------|---------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C10H18O2/c1-4-6-10(11)12-8-5-7-9(2)3/h4,6,9H,5,7-8H2,1-3H3/b6-4+ |
| InchiKey:            | XVJNSXFXUYUWHCS-GQCTYLIASA-N                                              |
| Formula:             | C10H18O2                                                                  |
| SMILES:              | CC=CC(=O)OCCCC(C)C                                                        |
| Mol. weight [g/mol]: | 170.25                                                                    |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -122.82 | kJ/mol  | Joback Method  |
| hf            | -382.59 | kJ/mol  | Joback Method  |
| hfus          | 21.12   | kJ/mol  | Joback Method  |
| hvap          | 46.58   | kJ/mol  | Joback Method  |
| log10ws       | -2.48   |         | Crippen Method |
| logp          | 2.542   |         | Crippen Method |
| mcvol         | 154.900 | ml/mol  | McGowan Method |
| pc            | 2329.27 | kPa     | Joback Method  |
| ripol         | 1317.00 |         | NIST Webbook   |
| ripol         | 1304.00 |         | NIST Webbook   |
| ripol         | 1317.00 |         | NIST Webbook   |
| tb            | 508.21  | K       | Joback Method  |
| tc            | 691.70  | K       | Joback Method  |
| tf            | 254.54  | K       | Joback Method  |
| vc            | 0.594   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 350.97 | J/mol×K | 508.21          | Joback Method |
| cpg           | 365.14 | J/mol×K | 538.79          | Joback Method |
| cpg           | 378.69 | J/mol×K | 569.37          | Joback Method |
| cpg           | 391.64 | J/mol×K | 599.96          | Joback Method |
| cpg           | 404.00 | J/mol×K | 630.54          | Joback Method |
| cpg           | 415.79 | J/mol×K | 661.12          | Joback Method |
| cpg           | 427.02 | J/mol×K | 691.70          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0043796 | Pa×s | 254.54 | Joback Method |
| dvisc | 0.0017548 | Pa×s | 296.82 | Joback Method |
| dvisc | 0.0008832 | Pa×s | 339.10 | Joback Method |
| dvisc | 0.0005176 | Pa×s | 381.38 | Joback Method |
| dvisc | 0.0003375 | Pa×s | 423.65 | Joback Method |
| dvisc | 0.0002378 | Pa×s | 465.93 | Joback Method |
| dvisc | 0.0001776 | Pa×s | 508.21 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R216667&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R216667&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>ripol:</b>   | Polar retention indices                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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