

# 4-Bromo-2,5-dimethoxyphenethylamine

<b>Other names:</b>	Bromo-2,5-dimethoxyphenylethylamine 2-CB BDMPEA 2,5-Dimethoxy-4-bromophenethylamine Benzeneethanamine, 4-bromo-2,5-dimethoxy-
<b>Inchi:</b>	InChI=1S/C10H14BrNO2/c1-13-9-6-8(11)10(14-2)5-7(9)3-4-12/h5-6H,3-4,12H2,1-2H3
<b>InchiKey:</b>	YMHOBZXQZVXHBM-UHFFFAOYSA-N
<b>Formula:</b>	C10H14BrNO2
<b>SMILES:</b>	COc1cc(CCN)c(OC)cc1Br
<b>Mol. weight [g/mol]:</b>	260.13
<b>CAS:</b>	66142-81-2

## Physical Properties

Property code	Value	Unit	Source
gf	-12.39	kJ/mol	Joback Method
hf	-251.93	kJ/mol	Joback Method
hfus	27.39	kJ/mol	Joback Method
hvap	64.01	kJ/mol	Joback Method
log10ws	-3.11		Crippen Method
logp	1.968		Crippen Method
mcvol	167.220	ml/mol	McGowan Method
pc	3086.42	kPa	Joback Method
rinpol	1841.00		NIST Webbook
rinpol	1841.00		NIST Webbook
tb	653.35	K	Joback Method
tc	879.36	K	Joback Method
tf	453.96	K	Joback Method
vc	0.615	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	400.90	J/molxK	653.35	Joback Method
cpg	413.53	J/molxK	691.02	Joback Method

cpg	425.44	J/mol×K	728.69	Joback Method
cpg	436.62	J/mol×K	766.35	Joback Method
cpg	447.07	J/mol×K	804.02	Joback Method
cpg	456.80	J/mol×K	841.69	Joback Method
cpg	465.79	J/mol×K	879.36	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C66142812&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C66142812&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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