

# p-Menth-1-en-7,8-diol

<b>Other names:</b>	p-1-Menthen-7,8-diol 7-Hydroxyterpineol
<b>Inchi:</b>	InChI=1S/C10H18O2/c1-10(2,12)9-5-3-8(7-11)4-6-9/h3,9,11-12H,4-7H2,1-2H3
<b>InchiKey:</b>	XYKGEKWHBMLSGS-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O2
<b>SMILES:</b>	CC(C)(O)C1CC=C(CO)CC1
<b>Mol. weight [g/mol]:</b>	170.25

## Physical Properties

Property code	Value	Unit	Source
gf	-192.70	kJ/mol	Joback Method
hf	-462.31	kJ/mol	Joback Method
hfus	15.09	kJ/mol	Joback Method
hvap	71.30	kJ/mol	Joback Method
log10ws	-2.15		Crippen Method
logp	1.476		Crippen Method
mcvol	148.340	ml/mol	McGowan Method
pc	3269.04	kPa	Joback Method
ripol	1469.00		NIST Webbook
ripol	1431.00		NIST Webbook
ripol	1502.00		NIST Webbook
ripol	1469.00		NIST Webbook
ripol	1502.00		NIST Webbook
ripol	1431.00		NIST Webbook
ripol	2517.00		NIST Webbook
ripol	2519.00		NIST Webbook
ripol	2528.00		NIST Webbook
ripol	2526.00		NIST Webbook
ripol	2550.00		NIST Webbook
ripol	2528.00		NIST Webbook
ripol	2526.00		NIST Webbook
ripol	2528.00		NIST Webbook
tb	633.02	K	Joback Method
tc	821.85	K	Joback Method
tf	347.18	K	Joback Method
vc	0.541	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	412.62	J/molxK	633.02	Joback Method
cpg	470.54	J/molxK	790.38	Joback Method
cpg	460.32	J/molxK	758.91	Joback Method
cpg	449.45	J/molxK	727.44	Joback Method
cpg	437.90	J/molxK	695.96	Joback Method
cpg	425.64	J/molxK	664.49	Joback Method
cpg	480.16	J/molxK	821.85	Joback Method
dvisc	0.0000228	Paxs	633.02	Joback Method
dvisc	0.0000429	Paxs	585.38	Joback Method
dvisc	0.0000903	Paxs	537.74	Joback Method
dvisc	0.0002192	Paxs	490.10	Joback Method
dvisc	0.0006446	Paxs	442.46	Joback Method
dvisc	0.0024586	Paxs	394.82	Joback Method
dvisc	0.0135407	Paxs	347.18	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R184862&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R184862&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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