

# 2',6'-Dimethoxyacetophenone

<b>Other names:</b>	2,6-Dimethoxyacetophenone Ethanone, 1-(2,6-dimethoxyphenyl)- Acetophenone, 2',6'-dimethoxy- USAF K-2801 2',6'-Dihydroxyacetophenone, dimethyl ether 1-(2,6-Dimethoxyphenyl)ethanone 1-(2,6-dimethoxyphenyl)ethan-1-one
<b>Inchi:</b>	InChI=1S/C10H12O3/c1-7(11)10-8(12-2)5-4-6-9(10)13-3/h4-6H,1-3H3
<b>InchiKey:</b>	XEUGKOFTNAYMMX-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O3
<b>SMILES:</b>	COc1cccc(OC)c1C(C)=O
<b>Mol. weight [g/mol]:</b>	180.20
<b>CAS:</b>	2040-04-2

## Physical Properties

Property code	Value	Unit	Source
gf	-212.45	kJ/mol	Joback Method
hf	-413.16	kJ/mol	Joback Method
hfus	18.89	kJ/mol	Joback Method
hvap	53.02	kJ/mol	Joback Method
log10ws	-2.37		Crippen Method
logp	1.906		Crippen Method
mcvol	141.310	ml/mol	McGowan Method
pc	2928.17	kPa	Joback Method
rinpol	1468.70		NIST Webbook
rinpol	1480.00		NIST Webbook
rinpol	1480.00		NIST Webbook
tb	563.55	K	Joback Method
tc	776.08	K	Joback Method
tf	348.31	K	Joback Method
vc	0.529	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	323.48	J/mol×K	563.55	Joback Method
cpg	381.23	J/mol×K	740.66	Joback Method
cpg	370.92	J/mol×K	705.24	Joback Method
cpg	359.98	J/mol×K	669.82	Joback Method
cpg	348.42	J/mol×K	634.39	Joback Method
cpg	336.25	J/mol×K	598.97	Joback Method
cpg	390.91	J/mol×K	776.08	Joback Method
dvisc	0.0001686	Paxs	563.55	Joback Method
dvisc	0.0002064	Paxs	527.68	Joback Method
dvisc	0.0002601	Paxs	491.80	Joback Method
dvisc	0.0003399	Paxs	455.93	Joback Method
dvisc	0.0004650	Paxs	420.06	Joback Method
dvisc	0.0006746	Paxs	384.18	Joback Method
dvisc	0.0010565	Paxs	348.31	Joback Method

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2040042&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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