

Propanoic acid, 2-phenoxy-, methyl ester

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|-----------------------------|--|
| Other names: | Methyl 2-phenoxypropionate |
| Inchi: | InChI=1S/C10H12O3/c1-8(10(11)12-2)13-9-6-4-3-5-7-9/h3-8H,1-2H3 |
| InchiKey: | KIBRBMKSBBVQDMK-UHFFFAOYSA-N |
| Formula: | C10H12O3 |
| SMILES: | COC(=O)C(C)Oc1ccccc1 |
| Mol. weight [g/mol]: | 180.20 |
| CAS: | 2065-24-9 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -195.63 | kJ/mol | Joback Method |
| hf | -395.50 | kJ/mol | Joback Method |
| hfus | 16.15 | kJ/mol | Joback Method |
| hvap | 51.31 | kJ/mol | Joback Method |
| log10ws | -1.82 | | Crippen Method |
| logp | 1.627 | | Crippen Method |
| mcvol | 141.310 | ml/mol | McGowan Method |
| pc | 3052.41 | kPa | Joback Method |
| rinpol | 1258.00 | | NIST Webbook |
| tb | 553.15 | K | Joback Method |
| tc | 768.17 | K | Joback Method |
| tf | 308.27 | K | Joback Method |
| vc | 0.523 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 324.75 | J/molxK | 553.15 | Joback Method |
| cpg | 338.43 | J/molxK | 588.99 | Joback Method |
| cpg | 351.37 | J/molxK | 624.82 | Joback Method |
| cpg | 363.58 | J/molxK | 660.66 | Joback Method |
| cpg | 375.04 | J/molxK | 696.50 | Joback Method |
| cpg | 385.78 | J/molxK | 732.34 | Joback Method |
| cpg | 395.79 | J/molxK | 768.17 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0022528 | Paxs | 308.27 | Joback Method |
| dvisc | 0.0011324 | Paxs | 349.08 | Joback Method |
| dvisc | 0.0006573 | Paxs | 389.90 | Joback Method |
| dvisc | 0.0004230 | Paxs | 430.71 | Joback Method |
| dvisc | 0.0002938 | Paxs | 471.52 | Joback Method |
| dvisc | 0.0002163 | Paxs | 512.34 | Joback Method |
| dvisc | 0.0001666 | Paxs | 553.15 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C2065249&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/85-810-8/Propanoic-acid-2-phenoxy-methyl-ester.pdf>

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