1-Butanone, 1-(4-hydroxyphenyl)-

Other names: 1-(4-hydroxyphenyl)-1-butanone

1-(4-hydroxyphenyl)butan-1-one

4'-hydroxybutyrophenone

4-Butanoylphenol4-Butyrylphenol

4-Hydroxybutyrophenone Butyrophenone, 4'-hydroxy-

NSC 17548 p-Butyrylphenol

p-Hydroxyphenyl propyl ketone

p-hydroxybutyrophenone

InChi: InChi=1S/C10H12O2/c1-2-3-10(12)8-4-6-9(11)7-5-8/h4-7,11H,2-3H2,1H3

InchiKey: GFBLPULLSAPXDC-UHFFFAOYSA-N

Formula: C10H12O2

SMILES: CCCC(=O)c1ccc(O)cc1

Mol. weight [g/mol]: 164.20 **CAS:** 1009-11-6

Physical Properties

Property code	Value	Unit	Source	
gf	-137.81	kJ/mol	Joback Method	
hf	-303.09	kJ/mol	Joback Method	
hfus	23.08	kJ/mol	Joback Method	
hvap	59.89	kJ/mol	Joback Method	
log10ws	-2.52		Crippen Method	
logp	2.375		Crippen Method	
mcvol	135.440	ml/mol	McGowan Method	
рс	3740.80	kPa	Joback Method	
tb	589.37	K	Joback Method	
tc	815.81	K	Joback Method	
tf	364.70	К	Thermochemistry of 4-HOC6H4COR (R = H, CH3, C2H5, n-C3H7, n-C4H9, n-C5H11, and n-C6H13) compounds	
VC	0.460	m3/kmol	Joback Method	

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	327.20	J/mol×K	589.37	Joback Method
cpg	339.55	J/mol×K	627.11	Joback Method
cpg	351.04	J/mol×K	664.85	Joback Method
cpg	361.74	J/mol×K	702.59	Joback Method
cpg	371.74	J/mol×K	740.33	Joback Method
cpg	381.12	J/mol×K	778.07	Joback Method
cpg	389.96	J/mol×K	815.81	Joback Method
dvisc	0.0014890	Paxs	390.53	Joback Method
dvisc	0.0006624	Paxs	423.67	Joback Method
dvisc	0.0003314	Paxs	456.81	Joback Method
dvisc	0.0001821	Paxs	489.95	Joback Method
dvisc	0.0001079	Paxs	523.09	Joback Method
dvisc	0.0000681	Paxs	556.23	Joback Method
dvisc	0.0000452	Paxs	589.37	Joback Method

Sources

McGowan Method: http://link.springer.com/article/10.1007/BF02311772

NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C1009116&Units=SI

Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Thermochemistry of 4-HOC6H4COR (R = H, CH3, C2H5, n-C3H7, n-C4H9, dog=nk-Method=c6H13) compounds: https://www.doi.org/10.1016/j.jct.2016.09.026 https://en.wikipedia.org/wiki/Joback_method

Legend

cpg: Ideal gas heat capacity

dvisc: Dynamic viscosity

gf: Standard Gibbs free energy of formationhf: Enthalpy of formation at standard conditionshfus: Enthalpy of fusion at standard conditions

hvap: Enthalpy of vaporization at standard conditions

log10ws: Log10 of Water solubility in mol/l

logp: Octanol/Water partition coefficient mcvol: McGowan's characteristic volume

pc: Critical Pressure

tb: Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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