

# Butanoic acid, phenyl ester

<b>Other names:</b>	Butyric acid, phenyl ester Phenyl butyrate
<b>Inchi:</b>	InChI=1S/C10H12O2/c1-2-6-10(11)12-9-7-4-3-5-8-9/h3-5,7-8H,2,6H2,1H3
<b>InchiKey:</b>	IGVPBCZDHMIOJH-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O2
<b>SMILES:</b>	CCCC(=O)Oc1ccccc1
<b>Mol. weight [g/mol]:</b>	164.20
<b>CAS:</b>	4346-18-3

## Physical Properties

Property code	Value	Unit	Source
gf	-88.19	kJ/mol	Joback Method
hf	-258.00	kJ/mol	Joback Method
hfus	18.48	kJ/mol	Joback Method
hvap	49.29	kJ/mol	Joback Method
log10ws	-2.62		Crippen Method
logp	2.392		Crippen Method
mvol	135.440	ml/mol	McGowan Method
pc	3082.99	kPa	Joback Method
rinpol	1245.00		NIST Webbook
tb	500.65 ± 1.00	K	NIST Webbook
tc	743.85	K	Joback Method
tf	301.04	K	Joback Method
vc	0.511	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	300.47	J/mol×K	531.17	Joback Method
cpg	361.01	J/mol×K	708.40	Joback Method
cpg	350.35	J/mol×K	672.95	Joback Method
cpg	338.99	J/mol×K	637.51	Joback Method
cpg	326.90	J/mol×K	602.06	Joback Method
cpg	314.07	J/mol×K	566.62	Joback Method

cpg	370.98	J/mol×K	743.85	Joback Method
dvisc	0.0002164	Paxs	531.17	Joback Method
dvisc	0.0002757	Paxs	492.81	Joback Method
dvisc	0.0003660	Paxs	454.46	Joback Method
dvisc	0.0005118	Paxs	416.11	Joback Method
dvisc	0.0007662	Paxs	377.75	Joback Method
dvisc	0.0012567	Paxs	339.39	Joback Method
dvisc	0.0023380	Paxs	301.04	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4346183&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4346183&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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