

# 3-Methylbenzyl acetate

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | m-Methylbenzyl acetate<br>Benzenemethanol, 3-methyl-, acetate<br>Acetic acid, (3-methylphenyl)methyl ester |
| <b>Inchi:</b>               | InChI=1S/C10H12O2/c1-8-4-3-5-10(6-8)7-12-9(2)11/h3-6H,7H2,1-2H3  |
| <b>InchiKey:</b>            | QPTQLFCBVFKFLY-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C10H12O2   |
| <b>SMILES:</b>              | CC(=O)OCc1cccc(C)c1  |
| <b>Mol. weight [g/mol]:</b> | 164.20   |
| <b>CAS:</b>                 | 17369-57-2   |

## Physical Properties

| Property code | Value   | Unit                 | Source         |
|---------------|---------|----------------------|----------------|
| gf            | -97.82  | kJ/mol               | Joback Method  |
| hf            | -269.47 | kJ/mol               | Joback Method  |
| hfus          | 18.10   | kJ/mol               | Joback Method  |
| hvap          | 49.95   | kJ/mol               | Joback Method  |
| log10ws       | -2.53   |                      | Crippen Method |
| logp          | 2.058   |                      | Crippen Method |
| mcvol         | 135.440 | ml/mol               | McGowan Method |
| pc            | 3032.27 | kPa                  | Joback Method  |
| rinsol        | 1283.00 |                      | NIST Webbook   |
| rinsol        | 1283.00 |                      | NIST Webbook   |
| tb            | 536.15  | K                    | Joback Method  |
| tc            | 749.95  | K                    | Joback Method  |
| tf            | 313.56  | K                    | Joback Method  |
| vc            | 0.511   | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 300.27 | J/mol×K | 536.15          | Joback Method |
| cpg           | 313.54 | J/mol×K | 571.78          | Joback Method |
| cpg           | 326.10 | J/mol×K | 607.42          | Joback Method |
| cpg           | 337.97 | J/mol×K | 643.05          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 349.15    | J/molxK | 678.68 | Joback Method |
| cpg   | 359.66    | J/molxK | 714.31 | Joback Method |
| cpg   | 369.52    | J/molxK | 749.95 | Joback Method |
| dvisc | 0.0017515 | Paxs    | 313.56 | Joback Method |
| dvisc | 0.0010200 | Paxs    | 350.66 | Joback Method |
| dvisc | 0.0006587 | Paxs    | 387.76 | Joback Method |
| dvisc | 0.0004592 | Paxs    | 424.86 | Joback Method |
| dvisc | 0.0003392 | Paxs    | 461.95 | Joback Method |
| dvisc | 0.0002621 | Paxs    | 499.05 | Joback Method |
| dvisc | 0.0002099 | Paxs    | 536.15 | Joback Method |

## Sources

|                        |   |
|------------------------|---|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17369572&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17369572&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>   |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

Latest version available from:

<https://www.cheméo.com/cid/85-856-8/3-Methylbenzyl-acetate.pdf>

Generated by Cheméo on 2024-04-25 21:06:19.837736305 +0000 UTC m=+16368428.758313618.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.