

# 1,3-Dioxolane, 2-methyl-2-phenyl-

Other names:	Acetophenone ethylene ketal 2-Methyl-2-phenyl-1,3-dioxolane
Inchi:	InChI=1S/C10H12O2/c1-10(11-7-8-12-10)9-5-3-2-4-6-9/h2-6H,7-8H2,1H3
InchiKey:	GAOQAPRPYXWTFK-UHFFFAOYSA-N
Formula:	C10H12O2
SMILES:	CC1(c2ccccc2)OCCO1
Mol. weight [g/mol]:	164.20
CAS:	3674-77-9

## Physical Properties

Property code	Value	Unit	Source
chs	-5306.30 ± 2.00	kJ/mol	NIST Webbook
gf	4.55	kJ/mol	Joback Method
hf	-261.90 ± 2.20	kJ/mol	NIST Webbook
hfs	-343.80 ± 2.10	kJ/mol	NIST Webbook
hfus	19.29	kJ/mol	Joback Method
hsub	81.90	kJ/mol	NIST Webbook
hsub	81.88 ± 0.50	kJ/mol	NIST Webbook
hvap	48.26	kJ/mol	Joback Method
log10ws	-1.84		Crippen Method
logp	1.906		Crippen Method
mcvol	128.880	ml/mol	McGowan Method
pc	3718.02	kPa	Joback Method
tb	524.30	K	Joback Method
tc	770.55	K	Joback Method
tf	316.82	K	Joback Method
vc	0.469	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.71	J/mol×K	524.30	Joback Method
cpg	318.41	J/mol×K	565.34	Joback Method
cpg	333.71	J/mol×K	606.38	Joback Method

cpg	347.79	J/mol×K	647.42	Joback Method
cpg	360.86	J/mol×K	688.47	Joback Method
cpg	373.10	J/mol×K	729.51	Joback Method
cpg	384.69	J/mol×K	770.55	Joback Method
hsubt	81.90 ± 0.50	kJ/mol	308.50	NIST Webbook

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3674779&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3674779&amp;Units=SI</a>

## Legend

chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfs:	Solid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hsub:	Enthalpy of sublimation at standard conditions
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/85-860-3/1-3-Dioxolane-2-methyl-2-phenyl.pdf>

Generated by Cheméo on 2024-04-20 04:56:53.388658936 +0000 UTC m=+15878262.309236255.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.