

# 1-(2,3-dihydro-5-benzofuranyl)-ethanone

<b>Inchi:</b>	InChI=1S/C10H10O2/c1-7(11)8-2-3-10-9(6-8)4-5-12-10/h2-3,6H,4-5H2,1H3
<b>InchiKey:</b>	MMVUJVASBDVNGJ-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O2
<b>SMILES:</b>	CC(=O)c1ccc2c(c1)CCO2
<b>Mol. weight [g/mol]:</b>	162.19
<b>CAS:</b>	90843-31-5

## Physical Properties

Property code	Value	Unit	Source
affp	902.60	kJ/mol	NIST Webbook
basg	870.70	kJ/mol	NIST Webbook
gf	-20.11	kJ/mol	Joback Method
hf	-187.58	kJ/mol	Joback Method
hfus	21.56	kJ/mol	Joback Method
hvap	52.93	kJ/mol	Joback Method
log10ws	-2.53		Crippen Method
logp	1.824		Crippen Method
mcvol	124.580	ml/mol	McGowan Method
pc	3628.97	kPa	Joback Method
tb	557.07	K	Joback Method
tc	789.22	K	Joback Method
tf	352.60	K	Joback Method
vc	0.472	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	288.14	J/molxK	557.07	Joback Method
cpg	301.05	J/molxK	595.76	Joback Method
cpg	313.03	J/molxK	634.45	Joback Method
cpg	324.12	J/molxK	673.14	Joback Method
cpg	334.41	J/molxK	711.84	Joback Method
cpg	343.96	J/molxK	750.53	Joback Method
cpg	352.83	J/molxK	789.22	Joback Method

dvisc	0.0021013	Paxs	352.60	Joback Method
dvisc	0.0014854	Paxs	386.68	Joback Method
dvisc	0.0011107	Paxs	420.76	Joback Method
dvisc	0.0008675	Paxs	454.83	Joback Method
dvisc	0.0007013	Paxs	488.91	Joback Method
dvisc	0.0005829	Paxs	522.99	Joback Method
dvisc	0.0004955	Paxs	557.07	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C90843315&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C90843315&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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