

1(3H)-Isobenzofuranone, 3-ethoxy-

Other names:	Phthalide, 3-ethoxy- 3-Ethoxyphthalide
Inchi:	InChI=1S/C10H10O3/c1-2-12-10-8-6-4-3-5-7(8)9(11)13-10/h3-6,10H,2H2,1H3
InchiKey:	VCBNFFOSUGMASJ-UHFFFAOYSA-N
Formula:	C10H10O3
SMILES:	CCOC1OC(=O)c2ccccc21
Mol. weight [g/mol]:	178.18
CAS:	16824-02-5

Physical Properties

Property code	Value	Unit	Source
gf	-116.86	kJ/mol	Joback Method
hf	-353.79	kJ/mol	Joback Method
hfus	22.12	kJ/mol	Joback Method
hvap	51.87	kJ/mol	Joback Method
log10ws	-2.45		Crippen Method
logp	1.892		Crippen Method
mcvol	130.450	ml/mol	McGowan Method
pc	3368.44	kPa	Joback Method
tb	583.79	K	Joback Method
tc	817.83	K	Joback Method
tf	376.36	K	Joback Method
vc	0.490	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	322.24	J/molxK	583.79	Joback Method
cpg	336.32	J/molxK	622.80	Joback Method
cpg	349.55	J/molxK	661.80	Joback Method
cpg	361.95	J/molxK	700.81	Joback Method
cpg	373.53	J/molxK	739.82	Joback Method
cpg	384.30	J/molxK	778.82	Joback Method
cpg	394.27	J/molxK	817.83	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C16824025&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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