

1-[(3-Trifluoromethyl)phenyl]propanol-1

Inchi:	InChI=1S/C10H11F3O/c1-2-9(14)7-4-3-5-8(6-7)10(11,12)13/h3-6,9,14H,2H2,1H3
InchiKey:	SZSJCUKIEPGUMF-UHFFFAOYSA-N
Formula:	C10H11F3O
SMILES:	CCC(O)c1cccc(C(F)(F)F)c1
Mol. weight [g/mol]:	204.19
CAS:	618-97-3

Physical Properties

Property code	Value	Unit	Source
gf	-584.75	kJ/mol	Joback Method
hf	-779.26	kJ/mol	Joback Method
hfus	17.70	kJ/mol	Joback Method
hvap	53.34	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	3.149		Crippen Method
mcvol	139.180	ml/mol	McGowan Method
pc	2835.36	kPa	Joback Method
tb	546.18	K	Joback Method
tc	727.80	K	Joback Method
tf	291.41	K	Joback Method
vc	0.543	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	339.99	J/mol×K	546.18	Joback Method
cpg	351.56	J/mol×K	576.45	Joback Method
cpg	362.42	J/mol×K	606.72	Joback Method
cpg	372.62	J/mol×K	636.99	Joback Method
cpg	382.19	J/mol×K	667.26	Joback Method
cpg	391.16	J/mol×K	697.53	Joback Method
cpg	399.57	J/mol×K	727.80	Joback Method

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	362.50 ± 1.50	K	0.70	NIST Webbook

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C618973&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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