

# 1-Propene, 3,3',3''-[methylidynetris(oxy)]tris-

<b>Other names:</b>	Orthoformic acid, triallyl ester Triallyl orthoformate Tris(allyloxy)methane 3,3',3''-[methylidynetris(oxy)]trispropene
<b>Inchi:</b>	InChI=1S/C10H16O3/c1-4-7-11-10(12-8-5-2)13-9-6-3/h4-6,10H,1-3,7-9H2
<b>InchiKey:</b>	OGRORDCCINVZKC-UHFFFAOYSA-N
<b>Formula:</b>	C10H16O3
<b>SMILES:</b>	C=CCOC(OCC=C)OCC=C
<b>Mol. weight [g/mol]:</b>	184.23
<b>CAS:</b>	16754-50-0

## Physical Properties

Property code	Value	Unit	Source
gf	-20.60	kJ/mol	Joback Method
hf	-275.38	kJ/mol	Joback Method
hfus	17.86	kJ/mol	Joback Method
hvap	42.69	kJ/mol	Joback Method
ie	9.80 ± 0.07	eV	NIST Webbook
log10ws	-1.93		Crippen Method
logp	1.878		Crippen Method
mcvol	156.470	ml/mol	McGowan Method
pc	2278.41	kPa	Joback Method
tb	485.06	K	Joback Method
tc	660.14	K	Joback Method
tf	248.87	K	Joback Method
vc	0.587	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.56	J/mol×K	485.06	Joback Method
cpg	403.31	J/mol×K	630.96	Joback Method
cpg	392.07	J/mol×K	601.78	Joback Method
cpg	380.38	J/mol×K	572.60	Joback Method

cpg	368.23	J/mol×K	543.42	Joback Method
cpg	355.62	J/mol×K	514.24	Joback Method
cpg	414.09	J/mol×K	660.14	Joback Method
dvisc	0.0001384	Paxs	485.06	Joback Method
dvisc	0.0001805	Paxs	445.70	Joback Method
dvisc	0.0002480	Paxs	406.33	Joback Method
dvisc	0.0003647	Paxs	366.97	Joback Method
dvisc	0.0005884	Paxs	327.60	Joback Method
dvisc	0.0010817	Paxs	288.24	Joback Method
dvisc	0.0024111	Paxs	248.87	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16754500&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16754500&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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