

# S-(+)-5-(1-Hydroxy-1-methylethyl)-2-methyl-2-cycl

<b>Other names:</b>	5-(1-Hydroxy-1-methylethyl)-2-methyl-2-cyclohexen-1-one, (S)- D-8-Hydroxycarvotanacetone (S)-8-Hydroxy-p-menth-1-en-6-one Carvone hydrate Carvon hydrate
<b>Inchi:</b>	InChI=1S/C10H16O2/c1-7-4-5-8(6-9(7)11)10(2,3)12/h4,8,12H,5-6H2,1-3H3/t8-/m0/s1
<b>InchiKey:</b>	DJOOMNLGIUGRKD-QMMMGPBSA-N
<b>Formula:</b>	C10H16O2
<b>SMILES:</b>	CC1=CCC(C(C)(C)O)CC1=O
<b>Mol. weight [g/mol]:</b>	168.23
<b>CAS:</b>	60593-11-5

## Physical Properties

Property code	Value	Unit	Source
gf	-178.47	kJ/mol	Joback Method
hf	-447.78	kJ/mol	Joback Method
hfus	10.51	kJ/mol	Joback Method
hvap	58.87	kJ/mol	Joback Method
log10ws	-2.17		Crippen Method
logp	1.683		Crippen Method
mcvol	144.040	ml/mol	McGowan Method
pc	3065.95	kPa	Joback Method
rinpol	1424.00		NIST Webbook
ripol	1754.00		NIST Webbook
tb	608.66	K	Joback Method
tc	822.00	K	Joback Method
tf	354.58	K	Joback Method
vc	0.529	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	384.00	J/molxK	608.66	Joback Method
cpg	399.22	J/molxK	644.22	Joback Method

cpg	413.54	J/mol×K	679.77	Joback Method
cpg	426.99	J/mol×K	715.33	Joback Method
cpg	439.59	J/mol×K	750.88	Joback Method
cpg	451.34	J/mol×K	786.44	Joback Method
cpg	462.27	J/mol×K	822.00	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C60593115&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C60593115&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m cvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r inpol:</b>	Non-polar retention indices
<b>r ipol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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