

# 2,5-Di-O-acetyl-1,4-anhydro-3-O-methyl-D-xylitol

**Other names:** 2,5-O-diAcetyl-1,4-Anhydro-3-O-methyl-D-xylitol.

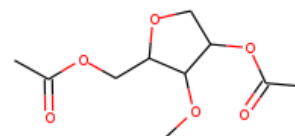
**InChI:** InChI=1S/C10H16O6/c1-6(11)14-4-8-10(13-3)9(5-15-8)16-7(2)12/h8-10H,4-5H2,1-3H3

**InChI Key:** SULWLWFNHXEWRP-UHFFFAOYSA-N

**Formula:** C10H16O6

**SMILES:** COC1C(COC(C)=O)OCC1OC(C)=O

**Molecular Weight:** 232.23



## Physical Properties

Property	Value	Unit	Source
$\Delta_f G^\circ$	-604.51	kJ/mol	Joback Method
$\Delta_f H^\circ_{\text{gas}}$	-983.75	kJ/mol	Joback Method
$\Delta_{\text{fus}} H^\circ$	32.47	kJ/mol	Joback Method
$\Delta_{\text{vap}} H^\circ$	62.72	kJ/mol	Joback Method
$\log P_{\text{oct/wat}}$	-0.11		Crippen Method
$P_c$	2505.01	kPa	Joback Method
$T_{\text{boil}}$	636.09	K	Joback Method
$T_c$	837.04	K	Joback Method
$T_{\text{fus}}$	398.00	K	Joback Method
$V_c$	0.62	m <sup>3</sup> /kg-mol	Joback Method

## Temperature Dependent Properties

Property	Value	Unit	Temperature (K)	Source
$C_{p,\text{gas}}$	464.89	J/mol×K	636.09	Joback Method
$\eta$	0.00	Paxs	636.09	Joback Method

## Sources

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**NIST Webbook:** [http://webbook.nist.gov/cgi/inchi/InChI=1S/C10H16O6/c1-6\(11\)14-4-8-10\(13-3\)9\(5-15-8\)16-7\(2\)12/h8-10H,4-5H2,1-3H3](http://webbook.nist.gov/cgi/inchi/InChI=1S/C10H16O6/c1-6(11)14-4-8-10(13-3)9(5-15-8)16-7(2)12/h8-10H,4-5H2,1-3H3)

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

$C_{p,gas}$ : Ideal gas heat capacity (J/molxK).

$\eta$ : Dynamic viscosity (Pa $\times$ s).

$\Delta_f G^\circ$ : Standard Gibbs free energy of formation (kJ/mol).

$\Delta_f H^\circ_{gas}$ : Enthalpy of formation at standard conditions (kJ/mol).

$\Delta_{fus} H^\circ$ : Enthalpy of fusion at standard conditions (kJ/mol).

$\Delta_{vap} H^\circ$ : Enthalpy of vaporization at standard conditions (kJ/mol).

$logP_{oct/wat}$ : Octanol/Water partition coefficient .

$P_c$ : Critical Pressure (kPa).

$T_{boil}$ : Normal Boiling Point Temperature (K).

$T_c$ : Critical Temperature (K).

$T_{fus}$ : Normal melting (fusion) point (K).

$V_c$ : Critical Volume (m<sup>3</sup>/kg-mol).

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