

# 2-Propenoic acid, 3-(2,3-dimethoxyphenyl)-, (E)-

Other names:

trans-2,3-Dimethoxycinnamic acid

(2E)-3-(2,3-Dimethoxyphenyl)-2-propenoic acid

Inchi:

InChI=1S/C11H12O4/c1-14-9-5-3-4-8(11(9)15-2)6-7-10(12)13/h3-7H,1-2H3,(H,12,13)/b7

InchiKey:

QAXPUWGAGVERSJ-VOTSOKGWSA-N

Formula:

C11H12O4

SMILES:

COc1cccc(C=CC(=O)O)c1OC

Mol. weight [g/mol]:

208.21

CAS:

7345-82-6

## Physical Properties

Property code	Value	Unit	Source
gf	-260.63	kJ/mol	Joback Method
hf	-468.81	kJ/mol	Joback Method
hfus	25.77	kJ/mol	Joback Method
hsub	141.00 ± 0.90	kJ/mol	NIST Webbook
hvap	71.88	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	1.802		Crippen Method
mcvol	156.970	ml/mol	McGowan Method
pc	3086.42	kPa	Joback Method
rinpol	1818.80		NIST Webbook
rinpol	1818.80		NIST Webbook
tb	682.77	K	Joback Method
tc	886.19	K	Joback Method
tf	415.32	K	Joback Method
vc	0.585	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	399.37	J/mol×K	682.77	Joback Method
cpg	446.99	J/mol×K	852.29	Joback Method
cpg	438.68	J/mol×K	818.39	Joback Method
cpg	429.78	J/mol×K	784.48	Joback Method

cpg	420.26	J/molxK	750.58	Joback Method
cpg	410.13	J/molxK	716.67	Joback Method
cpg	454.69	J/molxK	886.19	Joback Method
dvisc	0.0000349	Paxs	682.77	Joback Method
dvisc	0.0000498	Paxs	638.19	Joback Method
dvisc	0.0000748	Paxs	593.62	Joback Method
dvisc	0.0001202	Paxs	549.04	Joback Method
dvisc	0.0002098	Paxs	504.47	Joback Method
dvisc	0.0004082	Paxs	459.89	Joback Method
dvisc	0.0009159	Paxs	415.32	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7345826&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7345826&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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