

# Thiourea, N-(1,1-dimethylethyl)-N'-phenyl-

<b>Other names:</b>	Urea, 1-tert-butyl-3-phenyl-2-thio-
<b>Inchi:</b>	InChI=1S/C11H16N2S/c1-11(2,3)13-10(14)12-9-7-5-4-6-8-9/h4-8H,1-3H3,(H2,12,13,14)
<b>InchiKey:</b>	JHMCTNQTJNGVRX-UHFFFAOYSA-N
<b>Formula:</b>	C11H16N2S
<b>SMILES:</b>	CC(C)(C)NC(S)=Nc1ccccc1
<b>Mol. weight [g/mol]:</b>	208.32
<b>CAS:</b>	14327-04-9

## Physical Properties

Property code	Value	Unit	Source
hf	121.79	kJ/mol	Joback Method
hvap	57.63	kJ/mol	Joback Method
ie	7.85 ± 0.05	eV	NIST Webbook
log10ws	-3.66		Crippen Method
logp	2.992		Crippen Method
mcvol	174.100	ml/mol	McGowan Method
pc	2637.96	kPa	Joback Method
tb	664.12	K	Joback Method
tc	920.22	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14327049&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14327049&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

hf: Enthalpy of formation at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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