

1-tert-Butoxy-2-propanol, heptafluorobutyrate

Inchi:

InchiKey:

Formula:

SMILES:

Mol. weight [g/mol]:

InChI=1S/C11H15F7O3/c1-6(5-20-8(2,3)4)21-7(19)9(12,13)10(14,15)11(16,17)18/h6H,5H,7H,8H,9H,10H,11H,12H,13H,14H,15H,16H,17H,18H,19H,20H,21H

TTXDLKZACJODPX-UHFFFAOYSA-N

C11H15F7O3

CC(COC(C)(C)C)OC(=O)C(F)(F)C(F)(F)C(F)(F)F

328.22

Physical Properties

Property code	Value	Unit	Source
gf	-1651.93	kJ/mol	Joback Method
hf	-2060.44	kJ/mol	Joback Method
hfus	16.60	kJ/mol	Joback Method
hvap	40.36	kJ/mol	Joback Method
log10ws	-3.89		Crippen Method
logp	3.566		Crippen Method
mcvol	191.550	ml/mol	McGowan Method
pc	1631.17	kPa	Joback Method
rinpol	959.00		NIST Webbook
rinpol	959.00		NIST Webbook
tb	531.32	K	Joback Method
tc	689.39	K	Joback Method
tf	306.93	K	Joback Method
vc	0.769	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	511.48	J/mol×K	531.32	Joback Method
cpg	525.63	J/mol×K	557.67	Joback Method
cpg	538.95	J/mol×K	584.01	Joback Method
cpg	551.49	J/mol×K	610.36	Joback Method
cpg	563.28	J/mol×K	636.70	Joback Method
cpg	574.36	J/mol×K	663.05	Joback Method
cpg	584.76	J/mol×K	689.39	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U378347&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinsol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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