

# (S)-4-(1-Hydroxyallyl)phenyl acetate

<b>Inchi:</b>	InChI=1S/C11H12O3/c1-3-11(13)9-4-6-10(7-5-9)14-8(2)12/h3-7,11,13H,1H2,2H3/t11-/m
<b>InchiKey:</b>	GKYVDAMMLMMJGZ-LLVKDONJSA-N
<b>Formula:</b>	C11H12O3
<b>SMILES:</b>	C=CC(O)c1ccc(OC(C)=O)cc1
<b>Mol. weight [g/mol]:</b>	192.21
<b>CAS:</b>	343583-65-3

## Physical Properties

Property code	Value	Unit	Source
gf	-140.82	kJ/mol	Joback Method
hf	-322.19	kJ/mol	Joback Method
hfus	19.97	kJ/mol	Joback Method
hvap	67.80	kJ/mol	Joback Method
log10ws	-2.58		Crippen Method
logp	1.831		Crippen Method
mcvol	151.100	ml/mol	McGowan Method
pc	3191.93	kPa	Joback Method
rinpol	1550.20		NIST Webbook
rinpol	1550.20		NIST Webbook
tb	647.45	K	Joback Method
tc	851.46	K	Joback Method
tf	368.89	K	Joback Method
vc	0.561	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	377.46	J/molxK	647.45	Joback Method
cpg	388.70	J/molxK	681.45	Joback Method
cpg	399.26	J/molxK	715.45	Joback Method
cpg	409.15	J/molxK	749.45	Joback Method
cpg	418.40	J/molxK	783.46	Joback Method
cpg	427.03	J/molxK	817.46	Joback Method
cpg	435.05	J/molxK	851.46	Joback Method

dvisc	0.0028895	Paxs	368.89	Joback Method
dvisc	0.0010347	Paxs	415.32	Joback Method
dvisc	0.0004555	Paxs	461.74	Joback Method
dvisc	0.0002330	Paxs	508.17	Joback Method
dvisc	0.0001333	Paxs	554.60	Joback Method
dvisc	0.0000832	Paxs	601.02	Joback Method
dvisc	0.0000555	Paxs	647.45	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C343583653&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C343583653&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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