

# 2,5-Dimethoxycinnamic acid

<b>Other names:</b>	2-Propenoic acid, 3-(2,5-dimethoxyphenyl)- 3-(2,5-Dimethoxyphenyl)-2-propenoic acid 2',5'-dimethoxycinnamic acid trans-2,5-dimethoxycinnamic acid
<b>Inchi:</b>	InChI=1S/C11H12O4/c1-14-9-4-5-10(15-2)8(7-9)3-6-11(12)13/h3-7H,1-2H3,(H,12,13)/b6
<b>InchiKey:</b>	JPQWWJZORKTMIZ-ZZXKWWIFSA-N
<b>Formula:</b>	C11H12O4
<b>SMILES:</b>	COc1ccc(OC)c(C=CC(=O)O)c1
<b>Mol. weight [g/mol]:</b>	208.21
<b>CAS:</b>	10538-51-9

## Physical Properties

Property code	Value	Unit	Source
gf	-260.63	kJ/mol	Joback Method
hf	-468.81	kJ/mol	Joback Method
hfus	25.77	kJ/mol	Joback Method
hsub	138.80 ± 1.10	kJ/mol	NIST Webbook
hvap	71.88	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	1.802		Crippen Method
mcvol	156.970	ml/mol	McGowan Method
pc	3086.42	kPa	Joback Method
tb	682.77	K	Joback Method
tc	886.19	K	Joback Method
tf	415.32	K	Joback Method
vc	0.585	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	399.37	J/molxK	682.77	Joback Method
cpg	446.99	J/molxK	852.29	Joback Method
cpg	438.68	J/molxK	818.39	Joback Method
cpg	429.78	J/molxK	784.48	Joback Method

cpg	420.26	J/mol×K	750.58	Joback Method
cpg	410.13	J/mol×K	716.67	Joback Method
cpg	454.69	J/mol×K	886.19	Joback Method
dvisc	0.0000349	Paxs	682.77	Joback Method
dvisc	0.0000498	Paxs	638.19	Joback Method
dvisc	0.0000748	Paxs	593.62	Joback Method
dvisc	0.0001202	Paxs	549.04	Joback Method
dvisc	0.0002098	Paxs	504.47	Joback Method
dvisc	0.0004082	Paxs	459.89	Joback Method
dvisc	0.0009159	Paxs	415.32	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10538519&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10538519&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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