

# Propyl 2,4-dihydroxy-6-methylbenzoate

<b>Inchi:</b>	InChI=1S/C11H14O4/c1-3-4-15-11(14)10-7(2)5-8(12)6-9(10)13/h5-6,12-13H,3-4H2,1-2H
<b>InchiKey:</b>	WHFPMVZTSUOBEX-UHFFFAOYSA-N
<b>Formula:</b>	C11H14O4
<b>SMILES:</b>	CCCOC(=O)c1c(C)cc(O)cc1O
<b>Mol. weight [g/mol]:</b>	210.23
<b>CAS:</b>	31581-29-0

## Physical Properties

Property code	Value	Unit	Source
gf	-398.64	kJ/mol	Joback Method
hf	-644.73	kJ/mol	Joback Method
hfus	32.25	kJ/mol	Joback Method
hvap	78.20	kJ/mol	Joback Method
log10ws	-2.12		Crippen Method
logp	1.973		Crippen Method
mcvol	161.270	ml/mol	McGowan Method
pc	3838.78	kPa	Joback Method
rinpol	1786.20		NIST Webbook
rinpol	1786.20		NIST Webbook
tb	720.27	K	Joback Method
tc	948.77	K	Joback Method
tf	548.27	K	Joback Method
vc	0.499	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	443.44	J/molxK	720.27	Joback Method
cpg	495.18	J/molxK	910.69	Joback Method
cpg	485.49	J/molxK	872.61	Joback Method
cpg	475.58	J/molxK	834.52	Joback Method
cpg	465.34	J/molxK	796.44	Joback Method
cpg	454.66	J/molxK	758.35	Joback Method
cpg	504.75	J/molxK	948.77	Joback Method

dvisc	0.0000014	Paxs	720.27	Joback Method
dvisc	0.0000021	Paxs	691.60	Joback Method
dvisc	0.0000031	Paxs	662.94	Joback Method
dvisc	0.0000050	Paxs	634.27	Joback Method
dvisc	0.0000082	Paxs	605.60	Joback Method
dvisc	0.0000143	Paxs	576.94	Joback Method
dvisc	0.0000262	Paxs	548.27	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C31581290&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C31581290&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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