

# 1,2-Cyclohexanedicarboxylic acid, dodecyl 4-methoxyphenyl ester

Inchi:	InChI=1S/C27H42O5/c1-3-4-5-6-7-8-9-10-11-14-21-31-26(28)24-15-12-13-16-25(24)27(2)
InchiKey:	VSSFWSSBNIUSSY-UHFFFAOYSA-N
Formula:	C27H42O5
SMILES:	CCCCCCCCCCCCOC(=O)C1CCCCC1C(=O)Oc1ccc(OC)cc1
Mol. weight [g/mol]:	446.62

## Physical Properties

Property code	Value	Unit	Source
gf	-276.86	kJ/mol	Joback Method
hf	-963.39	kJ/mol	Joback Method
hfus	59.01	kJ/mol	Joback Method
hvap	99.48	kJ/mol	Joback Method
log10ws	-7.71		Crippen Method
logp	6.871		Crippen Method
mvol	377.420	ml/mol	McGowan Method
pc	948.50	kPa	Joback Method
rinpol	3334.00		NIST Webbook
rinpol	3334.00		NIST Webbook
tb	1038.70	K	Joback Method
tc	1271.71	K	Joback Method
tf	602.68	K	Joback Method
vc	1.438	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1333.49	J/molxK	1038.70	Joback Method
cpg	1348.88	J/molxK	1077.54	Joback Method
cpg	1362.13	J/molxK	1116.37	Joback Method
cpg	1373.27	J/molxK	1155.21	Joback Method
cpg	1382.35	J/molxK	1194.04	Joback Method
cpg	1389.40	J/molxK	1232.88	Joback Method
cpg	1394.47	J/molxK	1271.71	Joback Method
dvisc	0.0002399	Paxs	602.68	Joback Method

dvisc	0.0001268	Paxs	675.35	Joback Method
dvisc	0.0000759	Paxs	748.02	Joback Method
dvisc	0.0000497	Paxs	820.69	Joback Method
dvisc	0.0000349	Paxs	893.36	Joback Method
dvisc	0.0000258	Paxs	966.03	Joback Method
dvisc	0.0000200	Paxs	1038.70	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339676&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339676&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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