

Benzamide, N,N-didecyl-4-fluoro-

Inchi:	InChI=1S/C27H46FNO/c1-3-5-7-9-11-13-15-17-23-29(24-18-16-14-12-10-8-6-4-2)27(30)
InchiKey:	SEDQDGSOVLATCW-UHFFFAOYSA-N
Formula:	C27H46FNO
SMILES:	CCCCCCCCCN(CCCCCCCCCC)C(=O)c1ccc(F)cc1
Mol. weight [g/mol]:	419.66

Physical Properties

Property code	Value	Unit	Source
gf	66.29	kJ/mol	Joback Method
hf	-616.71	kJ/mol	Joback Method
hfus	67.04	kJ/mol	Joback Method
hvap	86.61	kJ/mol	Joback Method
log10ws	-9.48		Crippen Method
logp	8.549		Crippen Method
mvol	380.850	ml/mol	McGowan Method
pc	826.69	kPa	Joback Method
rinpol	2954.00		NIST Webbook
rinpol	2954.00		NIST Webbook
tb	914.40	K	Joback Method
tc	1119.72	K	Joback Method
tf	515.98	K	Joback Method
vc	1.482	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1275.42	J/molxK	914.40	Joback Method
cpg	1296.11	J/molxK	948.62	Joback Method
cpg	1315.54	J/molxK	982.84	Joback Method
cpg	1333.80	J/molxK	1017.06	Joback Method
cpg	1350.95	J/molxK	1051.28	Joback Method
cpg	1367.09	J/molxK	1085.50	Joback Method
cpg	1382.31	J/molxK	1119.72	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308312&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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