

Phthalic acid, dodecyl 4-methylthiophenyl ester

Inchi:	InChI=1S/C27H36O4S/c1-3-4-5-6-7-8-9-10-11-14-21-30-26(28)24-15-12-13-16-25(24)27
InchiKey:	VILBCMSDDKNXON-UHFFFAOYSA-N
Formula:	C27H36O4S
SMILES:	CCCCCCCCCCCCOC(=O)c1ccccc1C(=O)Oc1ccc(SC)cc1
Mol. weight [g/mol]:	456.64

Physical Properties

Property code	Value	Unit	Source
gf	-52.70	kJ/mol	Joback Method
hf	-598.22	kJ/mol	Joback Method
hfus	62.69	kJ/mol	Joback Method
hvap	106.70	kJ/mol	Joback Method
log10ws	-9.15		Crippen Method
logp	7.705		Crippen Method
mcvol	375.000	ml/mol	McGowan Method
pc	1061.71	kPa	Joback Method
rinpol	1487.00		NIST Webbook
rinpol	1487.00		NIST Webbook
tb	1101.84	K	Joback Method
tc	1348.97	K	Joback Method
tf	650.65	K	Joback Method
vc	1.433	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1247.91	J/mol×K	1101.84	Joback Method
cpg	1259.58	J/mol×K	1143.03	Joback Method
cpg	1269.43	J/mol×K	1184.22	Joback Method
cpg	1277.51	J/mol×K	1225.40	Joback Method
cpg	1283.90	J/mol×K	1266.59	Joback Method
cpg	1288.66	J/mol×K	1307.78	Joback Method
cpg	1291.84	J/mol×K	1348.97	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415621&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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