

# Isophthalic acid, 4-isopropylphenyl decyl ester

**Inchi:** InChI=1S/C27H36O4/c1-4-5-6-7-8-9-10-11-19-30-26(28)23-13-12-14-24(20-23)27(29)31  
**InchiKey:** XQUABHHGIKFHLE-UHFFFAOYSA-N  
**Formula:** C27H36O4  
**SMILES:** CCCCCCCCCOC(=O)c1cccc(C(=O)Oc2ccc(C(C)C)cc2)c1  
**Mol. weight [g/mol]:** 424.57

## Physical Properties

Property code	Value	Unit	Source
gf	-88.26	kJ/mol	Joback Method
hf	-645.37	kJ/mol	Joback Method
hfus	55.04	kJ/mol	Joback Method
hvap	99.50	kJ/mol	Joback Method
log10ws	-8.75		Crippen Method
logp	7.327		Crippen Method
mvol	358.650	ml/mol	McGowan Method
pc	1055.51	kPa	Joback Method
rinpol	3413.00		NIST Webbook
rinpol	3413.00		NIST Webbook
tb	1032.62	K	Joback Method
tc	1264.92	K	Joback Method
tf	601.25	K	Joback Method
vc	1.373	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1198.18	J/molxK	1032.62	Joback Method
cpg	1212.63	J/molxK	1071.34	Joback Method
cpg	1225.49	J/molxK	1110.05	Joback Method
cpg	1236.84	J/molxK	1148.77	Joback Method
cpg	1246.72	J/molxK	1187.49	Joback Method
cpg	1255.22	J/molxK	1226.20	Joback Method
cpg	1262.38	J/molxK	1264.92	Joback Method
dvisc	0.0002297	Paxs	601.25	Joback Method

dvisc	0.0001214	Paxs	673.14	Joback Method
dvisc	0.0000725	Paxs	745.04	Joback Method
dvisc	0.0000475	Paxs	816.93	Joback Method
dvisc	0.0000333	Paxs	888.83	Joback Method
dvisc	0.0000246	Paxs	960.72	Joback Method
dvisc	0.0000189	Paxs	1032.62	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344461&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344461&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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