

# Hexanoic acid, 3,5,5-trimethyl-, octadecyl ester

**Inchi:** InChI=1S/C27H54O2/c1-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-29-26(28)23-  
**InchiKey:** IRPUQPOQLCBARJ-UHFFFAOYSA-N  
**Formula:** C27H54O2  
**SMILES:** CCCCCCCCCCCCCCCCCOC(=O)CC(C)CC(C)(C)C  
**Mol. weight [g/mol]:** 410.72

## Physical Properties

Property code	Value	Unit	Source
gf	-57.06	kJ/mol	Joback Method
hf	-859.44	kJ/mol	Joback Method
hfus	57.54	kJ/mol	Joback Method
hvap	83.17	kJ/mol	Joback Method
log10ws	-9.50		Crippen Method
logp	9.253		Crippen Method
mvol	398.730	ml/mol	McGowan Method
pc	716.07	kPa	Joback Method
rinpol	2750.00		NIST Webbook
rinpol	2750.00		NIST Webbook
tb	889.78	K	Joback Method
tc	1090.17	K	Joback Method
tf	453.63	K	Joback Method
vc	1.554	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1350.73	J/molxK	889.78	Joback Method
cpg	1373.80	J/molxK	923.18	Joback Method
cpg	1395.51	J/molxK	956.58	Joback Method
cpg	1415.92	J/molxK	989.98	Joback Method
cpg	1435.12	J/molxK	1023.38	Joback Method
cpg	1453.17	J/molxK	1056.78	Joback Method
cpg	1470.13	J/molxK	1090.17	Joback Method
dvisc	0.0008352	Paxs	453.63	Joback Method

dvisc	0.0002878	Paxs	526.32	Joback Method
dvisc	0.0001284	Paxs	599.01	Joback Method
dvisc	0.0000682	Paxs	671.70	Joback Method
dvisc	0.0000410	Paxs	744.40	Joback Method
dvisc	0.0000270	Paxs	817.09	Joback Method
dvisc	0.0000190	Paxs	889.78	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406069&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406069&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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