

# Fumaric acid, 3,3-dimethylbut-2-yl heptadecyl ester

Inchi:	InChI=1S/C27H50O4/c1-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-23-30-25(28)21-22-
InchiKey:	OHGSCHFZBKXIW-QURGRASLSA-N
Formula:	C27H50O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C=CC(=O)OC(C)C(C)(C)C
Mol. weight [g/mol]:	438.68

## Physical Properties

Property code	Value	Unit	Source
gf	-210.76	kJ/mol	Joback Method
hf	-987.02	kJ/mol	Joback Method
hfus	60.52	kJ/mol	Joback Method
hvap	92.28	kJ/mol	Joback Method
log10ws	-8.57		Crippen Method
logp	7.935		Crippen Method
mvol	401.870	ml/mol	McGowan Method
pc	755.15	kPa	Joback Method
rinpol	2951.00		NIST Webbook
rinpol	2951.00		NIST Webbook
tb	970.23	K	Joback Method
tc	1191.32	K	Joback Method
tf	520.71	K	Joback Method
vc	1.558	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1381.79	J/molxK	970.23	Joback Method
cpg	1402.63	J/molxK	1007.08	Joback Method
cpg	1422.02	J/molxK	1043.93	Joback Method
cpg	1440.05	J/molxK	1080.78	Joback Method
cpg	1456.81	J/molxK	1117.62	Joback Method
cpg	1472.39	J/molxK	1154.47	Joback Method
cpg	1486.87	J/molxK	1191.32	Joback Method
dvisc	0.0003432	Paxs	520.71	Joback Method

dvisc	0.0001356	Paxs	595.63	Joback Method
dvisc	0.0000659	Paxs	670.55	Joback Method
dvisc	0.0000371	Paxs	745.47	Joback Method
dvisc	0.0000231	Paxs	820.39	Joback Method
dvisc	0.0000156	Paxs	895.31	Joback Method
dvisc	0.0000112	Paxs	970.23	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348716&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348716&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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