

# Benzene, 1,3-dimethoxy-2,4,5-trimethyl

<b>Inchi:</b>	InChI=1S/C11H16O2/c1-7-6-10(12-4)9(3)11(13-5)8(7)2/h6H,1-5H3
<b>InchiKey:</b>	MUVGPKFVOVVZDS-UHFFFAOYSA-N
<b>Formula:</b>	C11H16O2
<b>SMILES:</b>	COc1cc(C)c(C)c(OC)c1C
<b>Mol. weight [g/mol]:</b>	180.24

## Physical Properties

Property code	Value	Unit	Source
gf	-94.37	kJ/mol	Joback Method
hf	-344.16	kJ/mol	Joback Method
hfus	19.11	kJ/mol	Joback Method
hvap	49.82	kJ/mol	Joback Method
log10ws	-3.14		Crippen Method
logp	2.629		Crippen Method
mcvol	153.830	ml/mol	McGowan Method
pc	2410.00	kPa	Joback Method
rinpol	1390.00		NIST Webbook
rinpol	1390.00		NIST Webbook
tb	542.52	K	Joback Method
tc	745.65	K	Joback Method
tf	334.69	K	Joback Method
vc	0.580	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	351.47	J/molxK	542.52	Joback Method
cpg	365.56	J/molxK	576.37	Joback Method
cpg	379.13	J/molxK	610.23	Joback Method
cpg	392.16	J/molxK	644.08	Joback Method
cpg	404.65	J/molxK	677.94	Joback Method
cpg	416.57	J/molxK	711.79	Joback Method
cpg	427.92	J/molxK	745.65	Joback Method
dvisc	0.0006908	Paxs	334.69	Joback Method

dvisc	0.0004642	Paxs	369.33	Joback Method
dvisc	0.0003339	Paxs	403.97	Joback Method
dvisc	0.0002530	Paxs	438.61	Joback Method
dvisc	0.0001997	Paxs	473.24	Joback Method
dvisc	0.0001628	Paxs	507.88	Joback Method
dvisc	0.0001362	Paxs	542.52	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R142744&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R142744&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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<https://www.chemeo.com/cid/86-625-3/Benzene-1-3-dimethoxy-2-4-5-trimethyl.pdf>

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