

# (+)-N-Methylephedrine

<b>Other names:</b>	(1S,2R)-(+)-N-Methylephedrine Benzenemethanol, «alpha»-[1-(dimethylamino)ethyl]-, [S-(R*,S*)]- L-(+)-erythro-N-methylephedrine d-N-methylephedrine
<b>Inchi:</b>	InChI=1S/C11H17NO/c1-9(12(2)3)11(13)10-7-5-4-6-8-10/h4-9,11,13H,1-3H3
<b>InchiKey:</b>	FMCGSUUBYTWNDP-UHFFFAOYSA-N
<b>Formula:</b>	C11H17NO
<b>SMILES:</b>	CC(C(O)c1ccccc1)N(C)C
<b>Mol. weight [g/mol]:</b>	179.26
<b>CAS:</b>	42151-56-4

## Physical Properties

Property code	Value	Unit	Source
gf	123.23	kJ/mol	Joback Method
hf	-129.10	kJ/mol	Joback Method
hfus	18.35	kJ/mol	Joback Method
hvap	60.30	kJ/mol	Joback Method
log10ws	-1.93		Crippen Method
logp	1.670		Crippen Method
mcvol	157.940	ml/mol	McGowan Method
pc	2963.34	kPa	Joback Method
tb	581.50	K	Joback Method
tc	776.75	K	Joback Method
tf	303.44	K	Joback Method
vc	0.569	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	395.20	J/mol×K	581.50	Joback Method
cpg	409.59	J/mol×K	614.04	Joback Method
cpg	423.12	J/mol×K	646.58	Joback Method
cpg	435.84	J/mol×K	679.12	Joback Method
cpg	447.78	J/mol×K	711.67	Joback Method

cpg	458.99	J/mol×K	744.21	Joback Method
cpg	469.51	J/mol×K	776.75	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C42151564&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C42151564&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Solid-Liquid Equilibria of N-Methylephedrine Enantiomers and Their Mixtures in Three Chiral Solvents Distinguished by Chain Length:</b>	<a href="https://www.doi.org/10.1021/acs.jced.8b01245">https://www.doi.org/10.1021/acs.jced.8b01245</a>
<b>Solid-Liquid Equilibria of Chiral Ionic N-Methylephedrine Enantiomers and Their Mixtures in Three Chiral Solvents Distinguished by Chain Length:</b>	<a href="https://www.doi.org/10.1021/je1007839">https://www.doi.org/10.1021/je1007839</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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