

# Fumaric acid, di(3-(2-methoxyethyl)heptyl) ester

Inchi:	InChI=1S/C25H46O6/c1-5-7-10-22(15-19-28-3)12-9-18-30-24(26)13-14-25(27)31-21-17-
InchiKey:	NHSWDKGMJMKYTM-BUHFOSPRSA-N
Formula:	C25H46O6
SMILES:	CCCCC(CCCOC(=O)C=CC(=O)OCCC(CCCC)CCOC)CCOC
Mol. weight [g/mol]:	442.63

## Physical Properties

Property code	Value	Unit	Source
gf	-442.88	kJ/mol	Joback Method
hf	-1206.71	kJ/mol	Joback Method
hfus	61.61	kJ/mol	Joback Method
hvap	93.56	kJ/mol	Joback Method
log10ws	-5.56		Crippen Method
logp	5.485		Crippen Method
mvol	385.430	ml/mol	McGowan Method
pc	820.07	kPa	Joback Method
rinpol	2836.00		NIST Webbook
rinpol	2836.00		NIST Webbook
tb	972.10	K	Joback Method
tc	1195.25	K	Joback Method
tf	525.21	K	Joback Method
vc	1.488	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1317.22	J/molxK	972.10	Joback Method
cpg	1336.21	J/molxK	1009.29	Joback Method
cpg	1353.38	J/molxK	1046.48	Joback Method
cpg	1368.76	J/molxK	1083.68	Joback Method
cpg	1382.39	J/molxK	1120.87	Joback Method
cpg	1394.29	J/molxK	1158.06	Joback Method
cpg	1404.49	J/molxK	1195.25	Joback Method
dvisc	0.0002523	Paxs	525.21	Joback Method

dvisc	0.0001052	Paxs	599.69	Joback Method
dvisc	0.0000533	Paxs	674.17	Joback Method
dvisc	0.0000309	Paxs	748.65	Joback Method
dvisc	0.0000197	Paxs	823.14	Joback Method
dvisc	0.0000136	Paxs	897.62	Joback Method
dvisc	0.0000099	Paxs	972.10	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405911&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405911&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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