

# Succinic acid, 2-isopropoxyphenyl 2-biphenyl ester

Inchi:	InChI=1S/C25H24O5/c1-18(2)28-22-14-8-9-15-23(22)30-25(27)17-16-24(26)29-21-13-7-
InchiKey:	LEISVPTUWWMUIO-UHFFFAOYSA-N
Formula:	C25H24O5
SMILES:	CC(C)Oc1ccccc1OC(=O)CCC(=O)Oc1ccccc1-c1ccccc1
Mol. weight [g/mol]:	404.46

## Physical Properties

Property code	Value	Unit	Source
gf	-97.69	kJ/mol	Joback Method
hf	-499.78	kJ/mol	Joback Method
hfus	45.09	kJ/mol	Joback Method
hvap	99.73	kJ/mol	Joback Method
log10ws	-7.42		Crippen Method
logp	5.432		Crippen Method
mvol	312.580	ml/mol	McGowan Method
pc	1511.67	kPa	Joback Method
rinpol	3087.00		NIST Webbook
rinpol	3087.00		NIST Webbook
tb	1035.96	K	Joback Method
tc	1282.66	K	Joback Method
tf	627.36	K	Joback Method
vc	1.171	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	992.21	J/molxK	1035.96	Joback Method
cpg	1028.38	J/molxK	1241.54	Joback Method
cpg	1024.37	J/molxK	1200.43	Joback Method
cpg	1018.80	J/molxK	1159.31	Joback Method
cpg	1011.62	J/molxK	1118.19	Joback Method
cpg	1002.77	J/molxK	1077.08	Joback Method
cpg	1030.89	J/molxK	1282.66	Joback Method
dvisc	0.0000197	Paxs	1035.96	Joback Method

dvisc	0.0000250	Paxs	967.86	Joback Method
dvisc	0.0000331	Paxs	899.76	Joback Method
dvisc	0.0000457	Paxs	831.66	Joback Method
dvisc	0.0000669	Paxs	763.56	Joback Method
dvisc	0.0001056	Paxs	695.46	Joback Method
dvisc	0.0001841	Paxs	627.36	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357977&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357977&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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