

# Fumaric acid, 2-isopropoxyphenyl naphth-2-ylmethyl ester

<b>Inchi:</b>	InChI=1S/C24H22O5/c1-17(2)28-21-9-5-6-10-22(21)29-24(26)14-13-23(25)27-16-18-11-
<b>InchiKey:</b>	SYWNLBBWUDQTHV-BUHFOSPRSA-N
<b>Formula:</b>	C24H22O5
<b>SMILES:</b>	CC(C)Oc1cccc1OC(=O)C=CC(=O)OCc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	390.43

## Physical Properties

Property code	Value	Unit	Source
gf	-31.65	kJ/mol	Joback Method
hf	-407.38	kJ/mol	Joback Method
hfus	45.68	kJ/mol	Joback Method
hvap	96.83	kJ/mol	Joback Method
log10ws	-6.74		Crippen Method
logp	4.832		Crippen Method
mvol	298.490	ml/mol	McGowan Method
pc	1587.28	kPa	Joback Method
rinpol	3097.00		NIST Webbook
rinpol	3097.00		NIST Webbook
tb	1009.54	K	Joback Method
tc	1252.90	K	Joback Method
tf	617.29	K	Joback Method
vc	1.125	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	930.11	J/molxK	1009.54	Joback Method
cpg	942.15	J/molxK	1050.10	Joback Method
cpg	952.99	J/molxK	1090.66	Joback Method
cpg	962.71	J/molxK	1131.22	Joback Method
cpg	971.40	J/molxK	1171.78	Joback Method
cpg	979.14	J/molxK	1212.34	Joback Method
cpg	986.04	J/molxK	1252.90	Joback Method
dvisc	0.0002884	Paxs	617.29	Joback Method

dvisc	0.0001781	Paxs	682.67	Joback Method
dvisc	0.0001196	Paxs	748.04	Joback Method
dvisc	0.0000856	Paxs	813.41	Joback Method
dvisc	0.0000644	Paxs	878.79	Joback Method
dvisc	0.0000504	Paxs	944.16	Joback Method
dvisc	0.0000407	Paxs	1009.54	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405717&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405717&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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