

# Fumaric acid, 1-phenylprop-1-yl naphth-2-ylmethyl ester

**Inchi:** InChI=1S/C24H22O4/c1-2-22(20-9-4-3-5-10-20)28-24(26)15-14-23(25)27-17-18-12-13-1  
**InchiKey:** NESIGXKNEXPZHS-CCEZHUSRSA-N  
**Formula:** C24H22O4  
**SMILES:** CCC(OC(=O)C=CC(=O)OCc1ccc2ccccc2c1)c1ccccc1  
**Mol. weight [g/mol]:** 374.43

## Physical Properties

Property code	Value	Unit	Source
gf	82.98	kJ/mol	Joback Method
hf	-263.69	kJ/mol	Joback Method
hfus	44.88	kJ/mol	Joback Method
hvap	93.75	kJ/mol	Joback Method
log10ws	-6.74		Crippen Method
logp	5.134		Crippen Method
mcvol	292.620	ml/mol	McGowan Method
pc	1628.54	kPa	Joback Method
rinpol	3072.00		NIST Webbook
rinpol	3072.00		NIST Webbook
tb	982.14	K	Joback Method
tc	1225.18	K	Joback Method
tf	582.54	K	Joback Method
vc	1.107	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	905.70	J/molxK	982.14	Joback Method
cpg	918.84	J/molxK	1022.65	Joback Method
cpg	930.93	J/molxK	1063.15	Joback Method
cpg	942.10	J/molxK	1103.66	Joback Method
cpg	952.46	J/molxK	1144.17	Joback Method
cpg	962.14	J/molxK	1184.67	Joback Method
cpg	971.24	J/molxK	1225.18	Joback Method
dvisc	0.0004430	Paxs	582.54	Joback Method

dvisc	0.0002594	Paxs	649.14	Joback Method
dvisc	0.0001678	Paxs	715.74	Joback Method
dvisc	0.0001169	Paxs	782.34	Joback Method
dvisc	0.0000862	Paxs	848.94	Joback Method
dvisc	0.0000664	Paxs	915.54	Joback Method
dvisc	0.0000530	Paxs	982.14	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405907&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405907&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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