

# 1,2-Cyclohexanedicarboxylic acid, decyl isobutyl ester

Inchi:	InChI=1S/C22H40O4/c1-4-5-6-7-8-9-10-13-16-25-21(23)19-14-11-12-15-20(19)22(24)26
InchiKey:	XTLZHMNYSYCTM-UHFFFAOYSA-N
Formula:	C22H40O4
SMILES:	CCCCCCCCCOC(=O)C1CCCCC1C(=O)OCC(C)C
Mol. weight [g/mol]:	368.55

## Physical Properties

Property code	Value	Unit	Source
gf	-319.18	kJ/mol	Joback Method
hf	-958.31	kJ/mol	Joback Method
hfus	47.69	kJ/mol	Joback Method
hvap	82.61	kJ/mol	Joback Method
log10ws	-5.93		Crippen Method
logp	5.676		Crippen Method
mvol	324.860	ml/mol	McGowan Method
pc	1069.36	kPa	Joback Method
rinpol	2439.00		NIST Webbook
rinpol	2439.00		NIST Webbook
tb	869.78	K	Joback Method
tc	1069.70	K	Joback Method
tf	470.16	K	Joback Method
vc	1.242	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1093.46	J/molxK	869.78	Joback Method
cpg	1176.80	J/molxK	1036.38	Joback Method
cpg	1162.98	J/molxK	1003.06	Joback Method
cpg	1147.76	J/molxK	969.74	Joback Method
cpg	1131.11	J/molxK	936.42	Joback Method
cpg	1113.02	J/molxK	903.10	Joback Method
cpg	1189.25	J/molxK	1069.70	Joback Method
dvisc	0.0000488	Paxs	869.78	Joback Method

dvisc	0.0000652	Paxs	803.18	Joback Method
dvisc	0.0000917	Paxs	736.57	Joback Method
dvisc	0.0001381	Paxs	669.97	Joback Method
dvisc	0.0002276	Paxs	603.37	Joback Method
dvisc	0.0004247	Paxs	536.76	Joback Method
dvisc	0.0009457	Paxs	470.16	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339431&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339431&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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